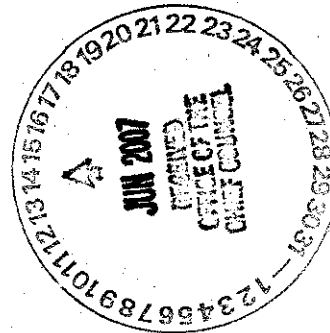


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STATE WATER RESOURCES CONTROL BOARD

OF THE STATE OF CALIFORNIA

In the Matter of the Petition of  
CHEVRON PRODUCTS COMPANY,  
CONOCOPHILLIPS COMPANY,  
EXXONMOBIL OIL CORPORATION, BP  
WEST COAST PRODUCTS, LLC, and  
WESTERN STATES PETROLEUM  
ASSOCIATION  
Request for Technical Reports, California  
Regional Water Quality Control Board, Los  
Angeles Region  
California Water Code § 13267

No.

VERIFIED PETITION FOR REVIEW  
AND REQUEST FOR HEARING

Chevron Products Company ("Chevron"), ConocoPhillips Company  
("ConocoPhillips"), ExxonMobil Oil Corporation ("ExxonMobil"), and BP West Coast  
Products, LLC (collectively "Refinery Petitioners"), and the Western States Petroleum  
Association ("WSPA") (together with Refinery Petitioners, "Petitioners") hereby timely file  
this Verified Petition for Review and Request for Hearing.

1           1.       Petitioners' mailing addresses are as follows:

<i><b>Petitioner</b></i>	<i><b>Address</b></i>
Chevron	324 W. El Segundo Blvd. El Segundo, CA 90245
ConocoPhillips	1520 E. Sepulveda Blvd. Carson, CA 90745
ConocoPhillips	150 Pier A Street Wilmington, CA 90744
ExxonMobil Oil	3700 West 190th Street Torrance, CA 90509
BP	2350 E. 223 <sup>rd</sup> Street Carson, CA 90810
WSPA	1415 L Street Sacramento, CA 95814-3801

11           Refinery Petitioners own and operate petroleum refineries and related marine  
12 terminal facilities in the Los Angeles metropolitan area. WSPA is a trade association that  
13 represents companies engaged in the exploration and production, refining, marketing and  
14 transportation of crude oil and refined products throughout the western United States. Each  
15 of the Refinery Petitioners is an active member of WSPA, and works collaboratively with  
16 other member companies, through WSPA, to address common issues of concern to the  
17 industry. On behalf of its members, WSPA has been an active participant in the Los  
18 Angeles Regional Water Quality Control Board's ("Water Board") development of Total  
19 Maximum Daily Loads ("TMDLs") for water bodies in the Los Angeles region.

20           2.       Petitioners seek review of a Requirement Under California Water Code  
21 Section 13267 for Submittal of Technical Reports on the Fate and Transport of Metals  
22 Emitted From [their respective facilities], issued by the Interim Executive Officer of the  
23 Water Board on May 15, 2007 ("13267 Requirement Letter"). Copies of the Requirement  
24 Letters issued to the Refinery Petitioners are attached as Exhibits 1-5. Petitioners  
25 understand that approximately 25 other industrial facilities — out of the literally thousands  
26 of stationary sources in the Basin — received comparable directives from the Water Board.  
27 The 13267 Requirement Letters require Refinery Petitioners to provide estimates of the  
28

1 total mass of certain metals that are *emitted to the atmosphere*, along with a discussion of  
2 the estimation methodology, uncertainties and assumption used in the total mass  
3 calculation. In addition, Refinery Petitioners are required to discuss the fate and transport  
4 of these metals, *i.e.*, how much is discharged via direct or indirect deposition to specified  
5 water bodies or to any other watersheds in the Los Angeles region. The Water Board states  
6 that this information is required to assess the significance of atmospheric deposition as a  
7 source of metals discharged to water bodies which are listed under section 303(d) of the  
8 federal Clean Water Act ("CWA") and that are the subject of upcoming TMDLs.<sup>1</sup>

9       3. This Petition is filed pursuant to section 13320 of the Water Code, which  
10 authorizes any aggrieved person to petition the State Water Resources Control Board  
11 ("State Board") to review any action (or failure to act) by a regional board. *See* Water  
12 Code, § 13223 (actions of the regional board shall include actions by its executive officer  
13 pursuant to powers and duties delegated to him by the regional board).

14       4. Petitioners challenge the Water Board's issuance of the 13267 Requirement  
15 Letters on the grounds that such action is outside the scope of the Water Board's legal  
16 authority under Water Code section 13267 and thus in violation of law. Further, the action  
17 of the Water Board is arbitrary and capricious in that it was taken precipitately, without due  
18 consideration of the specific manner in which the requested fate and transport studies were  
19 to be conducted, the amount and quality (statistical validity) of the data that may be  
20 produced from the studies, or whether the results of independently designed and conducted  
21 studies by relatively few emitting facilities could even be used to draw useful or  
22 scientifically reliable conclusions. Most importantly, the Water Board's action ignores the

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24 <sup>1</sup> Petitioners note that there is not a perfect correlation between the water bodies  
25 identified in the 13267 Requirement Letters, the pollutants that are included on the State's  
26 2006 303(d) list for those water bodies, and the pollutants for which information is sought  
27 in the Requirement Letters. For example, Santa Monica Bay is *not* listed as impaired for  
28 mercury or zinc, and presumably TMDLs will not be established for either of these  
pollutants. Nevertheless, the 13267 Requirement Letter issued to Chevron (which  
discharges to Santa Monica Bay under an NPDES permit), requires submittal of  
information on emissions of mercury and zinc. The Water Board's purpose in asking for  
this information is thus unclear.

1 fact that road dust associated with vehicular travel has already been demonstrated to be the  
2 primary source of aerial deposition of metals to waters in the region.

3 5. Petitioners are aggrieved because they are being directed to take actions they  
4 believe to be unauthorized and of no scientific merit. The Water Board's actions are taken  
5 without regard to the authority and function of the South Coast Air Quality Management  
6 District ("SCAQMD"), and threaten to interfere with the extensive regulatory framework  
7 that already exists around the estimation and control of emissions to the atmosphere.

8 6. Petitioners request that the State Board grant the relief requested in this  
9 Petition, as set forth in the Request for Relief.

10 7. Petitioners' statement of points and authorities in support of the issues raised  
11 by this Petition commences below.

12 8. A copy of this Petition is being sent by first-class mail to the Water Board,  
13 on June 14, 2007, to the attention of Ms. Deborah J. Smith, Interim Executive Officer.

14 9. The 13267 Requirement Letters were issued to Petitioners without any  
15 formal procedure or notice and opportunity to comment on the record. *See* 23 Cal. Code  
16 Regs., §§ 2050(a)(9), 2050(c).

17 10. Petitioners request a hearing to address the contentions in the Statement of  
18 Points and Authorities and reserve the right to present additional evidence. *See* 23 Cal.  
19 Code Regs., § 2050.6.

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1                                    **STATEMENT OF POINTS AND AUTHORITIES**

2            I.        **INTRODUCTION.**

3            A.        **Background.**

4            This Petition arises out of the Water Board's actions taken in direct response to a  
5   "petition" filed by the Natural Resources Defense Council ("NRDC") on March 12, 2007,  
6   demanding that requests for technical reports be issued to a number of stationary sources in  
7   the Los Angeles Basin, including petroleum refineries and related facilities owned and  
8   operated by the Refinery Petitioners, that emit various compounds to the air (namely lead,  
9   ammonia, copper, sulfates, zinc or mercury). Broadly proclaiming that the regional water  
10   quality control boards have the authority under the Porter-Cologne Water Quality Control  
11   Act ("Porter-Cologne Act") to investigate "any activity or factor that may affect water  
12   quality," and pointing to the Water Board's obligation to develop TMDLs for impaired  
13   water bodies in the Los Angeles area, NRDC blithely categorizes these emitting facilities as  
14   "dischargers" under the Water Code, and asserts they are subject to the full panoply of  
15   regulation under the statute, including the requirement to submit technical reports (Water  
16   Code, § 13267), to file Reports of Waste Discharge (Water Code, § 13260) and to obtain  
17   Waste Discharge Requirements (Water Code, § 13264) — all in respect of their emissions  
18   to the atmosphere.

19           Fundamental rules of statutory construction and interpretation, and over 50 years of  
20   administrative practice and precedent, compel a different result than the one reached by  
21   NRDC and by the Water Board in response to NRDC's "petition." Literalism aside,  
22   atmospheric emissions are not "discharges of waste" under the Water Code, and the water  
23   boards have no authority under the Water Code to regulate them as if they were.

24           Stationary sources are non-mobile sources of gaseous emissions to the atmosphere.  
25   The regulatory and permitting schemes that have been established by the U. S.  
26   Environmental Protection Agency ("EPA") and SCAQMD over a period of many years are  
27   extremely complex and establish an interwoven set of pervasive, stringent controls on  
28   virtually all but the smallest sources of atmospheric emissions. SCAQMD employs

1 hundreds of experienced air quality engineers and scientists who implement and oversee the  
2 air quality program in the South Coast and who are tasked to ensure that these regulations,  
3 and the permits that implement them, are based on sound science and good engineering  
4 practice. While Petitioners do not dispute that aerial deposition can be a source of pollutant  
5 loading to water bodies,<sup>2</sup> the Water Board's ad hoc foray into the world of air pollution  
6 control is without legal foundation and, if allowed to proceed along the path advocated by  
7 NRDC, would interfere significantly with the SCAQMD's regulatory function. While  
8 Water Board staff personnel are experienced in the area of water quality regulation, they  
9 have no experience in the vastly different field of air pollution control, as evidenced by the  
10 misguided demand for technical reports issued on May 15.

11 B. The Bay Area Air Deposition Study.

12 Both the Water Board and the NRDC petition point to the 13267 letter issued to the  
13 Bay Area petroleum refineries in February 2005, relating to mercury emissions, as  
14 precedent for the 13267 Requirement Letters being challenged herein. This reliance is  
15 misplaced, and more recent actions of the San Francisco Water Board under Water Code  
16 section 13267 are the subject of a pending Petition for Review and Request for Stay filed  
17 with the State Board on June 6, 2007. *See* SWRCB/OCC File A-1851 (hereafter "Bay Area  
18 Petition").

19 The February 2005 letter issued by the San Francisco Water Board requires the Bay  
20 Area refineries to conduct a study of the fate of mercury in air emissions from the  
21 refineries. This request was a negotiated compromise between San Francisco Water Board  
22 staff and the refiners, who disagreed over the proper interpretation of language in the

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23 <sup>2</sup> "Atmospheric deposition" is generally described as the process whereby pollutants  
24 are transferred from the air to the earth's surface.  
25 <http://www.epa.gov/gliindicators/air/airb.html> (last visited June 12, 2007). The terms "air  
26 deposition" and "aerial deposition" are also used to describe "atmospheric deposition."  
27 "Atmospheric deposition comes from *emissions* of air pollutants from natural and human-  
28 made (anthropogenic) sources." *See Frequently Asked Questions About Atmospheric  
Deposition: A Handbook for Watershed Managers*, U.S. EPA (September 2001), at 5  
(emphasis added). However, none of these terms is defined in the Porter-Cologne Act, the  
CWA or any corresponding regulations.

1 Mercury TMDL for San Francisco Bay that requires the Bay Area refiners to study “the  
2 environmental fate of mercury in crude oil” as a TMDL adaptive implementation measure.  
3 Staff interpreted this language as authorizing them to require the refiners to measure the  
4 amount of mercury in their incoming crudes and to conduct a refinery-wide mercury mass  
5 balance analysis. The Bay Area refiners disputed this interpretation, as well as the Water  
6 Board’s assertion of authority to investigate or regulate air emissions in the first instance.  
7 The Bay Area refiners ultimately agreed to conduct the air deposition study on a voluntary  
8 basis, effectively reaching a resolution of the dispute and avoiding a formal confrontation  
9 with staff. *See* Bay Area Petition, pp. 7-8. At no time did the Bay Area refineries concede  
10 that the Water Board had authority to require the air deposition study, nor has the Water  
11 Board’s authority in this regard ever been adjudicated.

12 On May 7, 2007, before the air deposition study could even be completed, the San  
13 Francisco Water Board issued an updated 13267 letter requesting information on the  
14 mercury content of crude oil processed by the refineries and requiring a refinery mercury  
15 mass balance analysis. The Bay Area refiners are seeking review of this action, including  
16 those aspects of the request which expanded the scope of the air deposition study beyond  
17 what was agreed upon in 2005. As a reflection of their good faith, and despite the fact that  
18 the request for a mass balance study violates the compromise reached in 2005, the Bay  
19 Area refineries are committed to completing the air deposition study they agreed to  
20 undertake in 2005, and are moving forward with that work, under a reservation of rights,  
21 pending the State Board’s review of the contested aspects of the new 13267 demand.  
22 Contrary to the assertions of the Water Board and NRDC, this voluntary commitment —  
23 reached as part of a compromise with the San Francisco Water Board — cannot be  
24 considered “precedent” that is in any way binding on sources in the Los Angeles region.

25 C. The Scope of the Water Board’s Authority Under Section 13267 Raises  
26 Significant Issues Appropriate for Review by the State Board.

27 For the reasons outlined above, and in consideration of the legal principles  
28 discussed below, the scope of the Water Board’s authority under Water Code section 13267

1 raises significant issues that are appropriate for review by the State Board. Unless this  
2 issue is resolved now, Refinery Petitioners — like the refineries in the Bay Area — are  
3 likely to be subject to increasingly far-reaching and technically unwarranted demands for  
4 information relating to air emissions, all based on the Water Board's untested assertion of  
5 authority under section 13267. In furtherance of the Water Board's basic authority under  
6 the Porter-Cologne Act, "[a] regional board, in establishing or reviewing any water quality  
7 control plan or waste discharge requirements, or in connection with any action relating to  
8 any plan or requirement *authorized by this division*, may investigate the quality of any  
9 waters of the state within its region." Water Code, § 13267(a) (emphasis added). In  
10 conducting such an investigation, the regional board may require technical or monitoring  
11 program reports from

12 any person who has *discharged, discharges*, or is  
13 suspected of *having discharged or discharging*, or who  
14 *proposes to discharge* waste within its region, or any  
15 citizen or domiciliary, or political agency or entity of this  
16 state who has *discharged, discharges*, or is *suspected of*  
*having discharged or discharging*, or who *proposes to*  
*discharge*, waste outside of its region that could affect the  
quality of waters within its region . . . .

17 *Id.* at §13267(b)(1) (emphasis added).

18 Based on the plain language of the statute, the authority of regional boards to  
19 conduct investigations pursuant to section 13267 is limited to activities which they are  
20 otherwise authorized to take under the statute, *i.e.*, those relating to "discharges" or  
21 "proposed discharges." The question that is squarely presented is whether the term  
22 "discharger" may reasonably be construed, within the context of the Porter-Cologne Act, to  
23 include a facility whose purported "discharges" are not to water bodies, but instead are  
24 gaseous emissions to the atmosphere. Petitioners respectfully submit that the answer to this  
25 question is "no." As discussed below, there is no support for such authority in the Water  
26 Code itself, or in any expressions of legislative intent during the enactment of the Porter-  
27 Cologne Act, or in relevant case law or State Board decisions. Moreover, the requirement  
28 that water boards develop TMDLs under section 303(d) of the Clean Water Act, 33 U.S.C.



1 § 1313(d), does not afford the boards the authority to regulate air emissions or to require  
2 companies to investigate air emissions as an antecedent to such regulation. The 13267  
3 Requirement Letters issued by the Water Board attempt to regulate activities beyond the  
4 scope of "discharges," and the Regional Board has thereby exceeded its statutory authority  
5 under the Water Code.

6 II. THE WATER BOARD HAS NO AUTHORITY UNDER THE PORTER-  
7 COLOGNE ACT TO REQUIRE STATIONARY SOURCES TO  
8 INVESTIGATE AIR EMISSIONS.

9 A. Refinery Petitioners' Status As NPDES Dischargers Does Not Authorize  
10 Issuance of the 13267 Requirement Letters.

11 As a preliminary matter, Refinery Petitioners' status as "dischargers" under the  
12 National Pollutant Discharge Elimination System ("NPDES") permit program is not  
13 relevant to whether the Water Board has authority to issue the contested 13267  
14 Requirement Letters to them. The NPDES permitting program is a vehicle by which the  
15 Water Board regulates "point source discharges to waters of the United States," in  
16 accordance with specific criteria and guidance set forth in the Clean Water Act and its  
17 implementing regulations. 33 U.S.C. § 1342. The NPDES program does not broaden the  
18 Water Board's authority under other provisions of state law (such as Water Code section  
19 13267), nor does it provide an independent source of authority that overrides express or  
20 implied limitations on the water boards' authority under state law. More specifically, the  
21 fact that a facility may hold an NPDES permit does not give the Water Board broader  
22 investigative or regulatory authority over that facility than it would otherwise have.<sup>3</sup>

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24 <sup>3</sup> It is worth noting that, while the Refinery Petitioners have NPDES permits, most of  
25 them discharge to Publicly Owned Treatment Works ("POTWs"), in accordance with the  
26 terms and conditions of local industrial discharge permits, and as such are not routinely  
27 regulated by the Water Board. With one exception, the refineries' NPDES permits are  
28 used primarily for storm water discharges and then only on an infrequent basis, given the  
low precipitation rates in the basin. For example, one of the refineries has not discharged  
under its NPDES permit since 2001. Another refinery has discharged on only two  
occasions during the past seven years.

1           B.     The Water Board's Authority is Defined and Limited by the Water Code.

2           The authority of an administrative agency is derived entirely from its enabling  
3 statute, and any agency action conducted outside the scope of such statutory authority is  
4 void. *Ass'n for Retarded Citizens v. Dep't of Developmental Serv.*, 38 Cal.3d 384, 391  
5 (1985) ("Administrative action that is not authorized by, or is inconsistent with, acts of the  
6 Legislature is void."); *City of South Pasadena v. Slater*, 56 F. Supp 2d 1106, 1144 (C.D.  
7 Cal. 1999) ("An administrative agency's failure to comply with the law invokes a public  
8 interest of the highest order: the interest in having government officials act in accordance  
9 with the law"). In this case, the Water Board is attempting to order private parties to  
10 investigate sources of air emissions, claiming such authority can be found in section 13267  
11 of the Water Code. As noted above, section 13267 applies to those who "discharge" or  
12 "propose to discharge" "waste" to waters of the State, and those terms circumscribe the  
13 scope of the Water Board's authority.<sup>4</sup> Emissions to the atmosphere are not "discharges"  
14 under the plain meaning of the word, and any interpretation that strains to reach a contrary  
15 conclusion ignores the long-settled rules of statutory construction as set forth below.

16           C.     Principles of Statutory Construction Require Consistent Interpretation of the  
17                   Law, in Accordance with Legislative Intent.

18           In order for the Water Board to lawfully investigate air emissions pursuant  
19 to section 13267, the term "discharge," as used in that section and elsewhere in the Water  
20 Code, would have to be defined to include emissions that cause atmospheric deposition to  
21 waters of the state. Despite its ubiquitous usage throughout the code, the term "discharge"  
22 is not defined in the statute except for purposes of Chapter 5.5 (Water Code, §§ 13370-  
23 13389) where it has the same meaning as "point source" under the federal CWA. Water  
24

25 \_\_\_\_\_  
26 <sup>4</sup> Other statutory provisions that grant investigatory authority to the water boards are  
27 similarly limited to "dischargers." See Water Code, § 13383(a) (The state board or a  
28 regional board may establish inspection requirements, "as authorized by Section 13160,  
13376, or 13377 or by subdivisions (b) and (c) of this section, for any person who  
*discharges, or proposes to discharge, to navigable waters . . .*").

1 Code, § 13373.<sup>5</sup> Where the legislature has not defined a word used in a statute (as is the  
2 case here), the ordinary rules of statutory construction apply, as discussed below.

3 In examining any statute, the State Board (as would a Court) must be guided by the  
4 well-established principle that its function (in this circumstance) is to “ascertain the intent  
5 of the lawmaker so as to effectuate the purpose of the law.” *People v. Pieters*, 52 Cal. 3d  
6 894, 898 (Cal. 1991); *Mir v. Charter Suburban Hospital*, 27 Cal. App. 4th 1471, 1482 (Cal.  
7 Ct. App. 1994). The State Board must determine such intent by first focusing on the words  
8 used by the legislature, giving them their ordinary meaning. *Mercer v. Department of*  
9 *Motor Vehicles*, 53 Cal. 3d 753, 763 (Cal. 1991). This is because “it is the language of the  
10 statute itself that has successfully braved the legislative gauntlet.” *Halbert’s Lumber, Inc.*  
11 *v. Lucky Stores, Inc.*, 6 Cal. App. 4th 1233, 1238 (Cal. Ct. App. 1992). In ascertaining the  
12 legislative purpose of the statute, the State Board must also consider “its objective, the evils  
13 which it is designed to prevent, **the character and context of the legislation in which the**  
14 **particular words appear**, the public policy enunciated or vindicated, the social history  
15 which attends it, **and the effect of the particular language on the entire statutory scheme.**”  
16 *Santa Barbara County Taxpayers Assn. v. County of Santa Barbara*, 194 Cal. App. 3d 674,  
17 680 (Cal. Ct. App. 1987) (emphasis added).

18 The Water Code was designed to regulate activities that cause releases or drainages  
19 of waste that flow into or enter waters of the state. See generally “*Recommended Changes*  
20 *in Water Quality Control, Final Report of the Study Panel to the California State Water*  
21 *Resources Control Board*” (March 1969) (the “Study Panel Report”) (proposing extensive  
22 changes to California water quality control laws). Sources of such “releases” and  
23 “drainages of waste” include point sources such as manufacturing and industrial facilities  
24 and POTWs, and non-point sources such as stormwater drainage and agricultural runoff.  
25 Waste from these sources can enter waterbodies directly by flowing into surface waters or  
26 infiltrating to groundwater, or indirectly by the application of waste to land that eventually

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27 <sup>5</sup> Plainly, emissions to the atmosphere are not “point sources” under the CWA.  
28

1 enters waters of the state through storm water runoff or migration of leachate.<sup>6</sup> Although  
2 few California courts have addressed the scope of "discharge" under the Water Code, the  
3 activities at issue in cases interpreting the term "waste" under the statute are particularly  
4 instructive as to what constitutes a "discharge." See *Sutter County v. Nicols* (1908) 152  
5 Cal. 688, 695-96 (the discharge of tailings and debris from a mine into rivers, constitutes a  
6 nuisance unless authorized by law); *People v. New Penn Mines, Inc.*, 212 Cal. App. 2d 667,  
7 673 (1963) (mine wastes, such as ore tailings and refinery run-off, are "industrial waste"  
8 within the Water Code); *Lake Madrone Water Dist. v. State Water Resources Control Bd.*,  
9 209 Cal.App.3d 163, 174 (1989) (concentrated sediment flushed from dam used to create  
10 recreational lake which clogged downstream creek constituted a "discharge of waste" for  
11 purposes of Water Code).

12 Under these basic rules, air "emissions" cannot reasonably be considered  
13 "discharges" under the Water Code. If for no other reason, this would extend the  
14 regulatory reach of the water boards (and the State Board) to virtually all industrial  
15 facilities in the state, given the thousands of water bodies in California. And, if  
16 "emissions" are "discharges," what would preclude vehicular exhaust from being classified  
17 as a "discharge"? Logically, if the Water Board believes it has authority to regulate  
18 sources of aerial deposition that have the potential to adversely affect water quality, then all  
19 motor vehicles — from huge fleets of diesel trucks to the private family car — are subject  
20 to the jurisdiction of the board. This is an absurd result that reveals the patent flaw in the  
21 Water Board's position.

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23 <sup>6</sup> "Waste" is defined under the Porter-Cologne Act to include "sewage and any and all  
24 other waste substances, liquid, solid, gaseous, or radioactive, associated with human  
25 habitation, or of human or animal origin, or from any producing, manufacturing, or  
26 processing operation, including waste placed within containers of whatever nature prior to,  
27 and for purposes of, disposal. Water Code, § 13050(d). Petitioners maintain that the  
28 reference to "gaseous" wastes in this section does not constitute a sufficient legal basis for  
issuance of the 13267 Requirement Letters. While the categories of "wastes" that are  
regulated under the Water Code are seemingly all-inclusive, they must in every case be  
"discharged" into waters of the state in order to fall within the scope of the Water Board's  
authority.

1 D. The Attorney General's Opinions on the Meaning of "Discharge" Do Not  
2 Support the Water Board's Extension of the Term to Air Emissions.

3 In the absence of a statutory or regulatory definition of "discharge," the California  
4 Attorney General has offered several legal opinions regarding the definition of "discharge."  
5 These opinions conclude that a "discharge" involves the addition of waste to waterbodies  
6 by either directly flowing into surface waters or groundwaters, or by the application of  
7 waste to land that eventually enters waters of the state during stormwater events. The  
8 Attorney General addressed this issue in 1956 in response to a request by the State Water  
9 Pollution Control Board for an opinion on whether numerous and varied industrial and  
10 agricultural activities "constitute a discharge of waste over which a regional water pollution  
11 control board has jurisdiction." 27 Op. Atty. Gen. 182 (1956). In concluding that all such  
12 activities were "discharges" subject to regional board jurisdiction, the Attorney General  
13 explained that:

14 the tests which control whether a discharge of waste under  
15 the jurisdiction of a regional water pollution control board  
16 is occurring are these. First, there must be a present  
17 discharge, that is, *a present flowing or issuing out*, of  
18 harmful material from the site of the particular operation  
19 into the waters of the State. The operation which produced  
20 the harmful material need not, however, be currently  
conducted. As was pointed out in 26 Ops. Cal. Atty. Gen.  
88, it is the *release or drainage* of that material *into* the  
waters of the State, and not the production of the material,  
which constitutes the discharge."

21 *Id.* (emphasis added). Factual settings in which the Attorney General has rendered opinions  
22 include: (1) *drainage, flow, or seepage* containing debris or earth from logging operations;<sup>7</sup>  
23 (2) *drainage, flow or seepage* containing garbage, ashes, mixed refuse, or solid industrial  
24 waste from dumps;<sup>8</sup> (3) *return irrigation or drainage water* from agricultural operations;<sup>9</sup>

25  
26 <sup>7</sup> 27 Ops. Atty. Gen. 182, 184 (1956).

27 <sup>8</sup> *Id.*

28 <sup>9</sup> *Id.*

1 (4) *run-off* from agricultural fields following routine application of pesticides;<sup>10</sup> (5) changes  
2 in the physical or chemical characteristics of receiving waters caused by extraction of  
3 minerals from a streambed;<sup>11</sup> and (6) dumping of earth moved from construction  
4 operations, or *drainage of wastewater* from construction sites.<sup>12</sup> None of these opinions  
5 makes any mention of air emissions associated with these activities. The common thread in  
6 all of these decisions is that all of the “waste”-producing activities which resulted in a  
7 “discharge” involved discharges to surface or ground waters or application to land — not to  
8 the air.

9 The traditional definition of “discharge,” as described above, is supported by the  
10 State Board regulations applicable to discharges of hazardous waste to land. *See* 23 Cal.  
11 Code Regs. § 2601, *et seq.* Under section 2601, the term “discharger” is defined as “any  
12 person who discharges waste which could affect the quality of waters of the state.” This  
13 same definition appears in the Title 27 regulations applicable to other classes of waste  
14 management units. *See* 27 Cal. Code Regs., § 20164. In the case of Class 1 (hazardous  
15 waste) units, the term “discharge” is further defined by reference to the state hazardous  
16 waste regulations as “the accidental or intentional *spilling, leaking, pumping, emitting,*  
17 *emptying or dumping* of hazardous waste into or on any land or water.” *See* 22 Cal. Code  
18 Regs., § 66260.10. “Emitting” in this context does not refer to gaseous emissions to the air.  
19 According to the doctrine of *ejusdem generis*, “when a statute contains a list or catalogue of  
20 items, a court should determine the meaning of each by reference to the others, giving  
21 preference to an interpretation that uniformly treats items similar in nature and scope.”  
22 *Ornela v. Randolph*, 4 Cal. 4th 1095, 1101 (1993). Air emissions — which involve the  
23 release of gas to the atmosphere — are in no way related to “spilling, leaking, emptying or  
24 dumping.” In fact, uncontained gaseous emissions such as those the Water Board seeks to  
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26 <sup>10</sup> 43 Ops. Atty. Gen. 302, 304 (1964).

27 <sup>11</sup> 32 Ops. Atty. Gen. 139, 140-41 (1958).

28 <sup>12</sup> 16 Ops. Atty. Gen. 125, 130-31 (1950).

1 investigate through the 13267 Requirement Letters are *not* subject to regulation under  
2 either state or federal hazardous waste laws despite the explicit reference to “emitting” in  
3 the definition of “discharge.” *In the Matter of: BP Chem. Am. Inc., Lima, Ohio*, 1991 WL  
4 208971 (E.P.A.); *see* 54 Fed. Reg. 50973 (December 11, 1989) (“EPA now believes our  
5 authority to identify or list a waste as hazardous under RCRA is limited to containerized or  
6 condensed gases (*i.e.*, section 1004(27) of RCRA excludes all other gases from the  
7 definition of solid wastes and thus cannot be considered hazardous wastes”); *see also*  
8 Health & Saf. Code, §§ 25110.11 and 25201.12 (excluding regulation of exhaust from the  
9 scope of the Hazardous Waste Control Law). Thus, it is clear for purposes of discharges of  
10 hazardous waste to land, that “discharges” are limited to the seepage or flow of waste to  
11 surface water, groundwater or to land. The same logically applies to discharges of other  
12 types of waste.

13 E. The Legislative History Does Not Support Inclusion of Air Emissions as  
14 Discharges under the Porter-Cologne Act.

15 It is a general rule of statutory construction that courts must give a statutory  
16 provision “a reasonable and common sense interpretation consistent with the apparent  
17 purpose and intention of the lawmakers . . . which upon application will result in wise  
18 policy rather than mischief or absurdity.” *DeYoung v. San Diego*, 147 Cal. App. 3d 11, 18  
19 (Cal. Ct. App. 1983), *overruled on other grounds in Yamaha Corp. of America v. State Bd.*  
20 *of Equalization*, 19 Cal. 4th 1, 15 (Cal. Cal. Ct. App. 1998). *See also Gustafson v. Alloyd*  
21 *Co., Inc.*, 513 U.S. 561, 575 (1995) (concluding that the phrase “any note” should not be  
22 interpreted to mean literally “any note,” but must be understood against the background of  
23 what Congress was attempting to accomplish in enacting the Securities Acts) (“We rely  
24 upon [the doctrine of *noscitur a sociis*] to avoid ascribing to one word a meaning so broad  
25 that it is inconsistent with its accompanying words, thus giving ‘unintended breadth to the  
26 Acts of Congress’”). Here, the Water Board essentially asserts that it has authority to  
27 regulate *any* act that may have an affect on water quality. This is an untenable position,  
28 contrary to legislative intent. *See Times Mirror Co. v. Superior Court*, 53 Cal. 3d 1325,

1 1334 n.7 (1991) (where a statute is *theoretically* capable of more than one construction, the  
2 Court must choose that which most comports with the intent of the legislature).

3 The Porter-Cologne Act became effective on January 1, 1970, replacing the Dickey  
4 Act as the primary water quality control law in California.<sup>13</sup> The Porter-Cologne Act was  
5 the first complete revision to California's water pollution control laws in two decades and  
6 was adopted, in part, to enable the State and regional water boards to better carry out the  
7 State's water quality objectives.<sup>14</sup> Prior to the adoption of the statute, the State Board  
8 commissioned a review of the existing water quality legislation. The Study Panel Report  
9 was submitted to the State Legislature, and "legislation in substantial conformance with the  
10 changes recommended by the Study Panel was signed by the Governor."<sup>15</sup> Although the  
11 Study Panel Report does not address the meaning of "discharge" under the Water Code, it  
12 provides insight into the types of activities that were considered sources of water pollution  
13 and thereby subject to regulation under the Act.

14 Significantly, the Study Panel was particularly concerned about the increase in *uses*  
15 *of water* that generate waste discharges:

16 During the last 20 years, there has been created a great deal  
17 more waste to be discharged – domestic and industrial  
18 waste, drainage from farmlands – all the side effects of  
19 more people, more prosperity, and more products. As a  
result, more than 80 percent of the water used nationwide  
by man has been previously used.

20 Study Panel Report, at 2. One of the primary concerns of the Study Panel was thus the  
21 need for a waste discharge permit program to control wastes entering waters of the state.  
22 The Study Panel recommended that point source discharges be controlled by establishing  
23 waste discharge requirements. *Id.* at 15. The Study Panel specifically recognized the  
24

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25 <sup>13</sup> Ronald B. Robie, *Water Pollution, An Affirmative Response by the State*  
26 *Legislature*, 1 PAC. L.J. 2 (1970).

26 <sup>14</sup> *Id.*

27 <sup>15</sup> Robie, *supra* note 13, at 4. The Study Panel Report includes "extensive notes  
28 intended to assist in determining legislative intent." *Id.* n.10.



1 regulatory challenges that come from non-point sources, noting "there are many  
2 uncontrollable pollutants which enter the waters of the state, such as runoff from urban and  
3 agricultural lands, for which provision[s] must be made." *Id.* However, even though the  
4 Study Panel recognized that many different sources affect water quality, air emissions were  
5 never considered as a source of water quality impairment through air deposition.

6 As noted above, the Attorney General squarely and frequently addressed the  
7 meaning of the word "discharge" under the Dickey Act. Accordingly, it comes as no  
8 surprise that the Legislature, in enacting the Porter-Cologne Act, did not feel obliged to  
9 include a definition of "discharge" in the new legislation. Since the Attorney General  
10 opinions pre-dated the Porter-Cologne Act (which lacks a definition of "discharge"), it can  
11 be inferred that the Legislature considered the meaning of "discharge" to be well  
12 understood and not different from that articulated by the Attorney General opinions.  
13 Moreover, it is evident that the scope of other definitions used in the Dickey Act remained  
14 unaltered following the Porter-Cologne Act. For example, under the Porter-Cologne Act,  
15 "the word 'waste' is intended to be as all inclusive as the Dickey Act definitions it replaces  
16 and, therefore, the opinions of the attorney general relating to discharges of 'sewage,'  
17 'other waste' and 'industrial waste' are still applicable.'" Robie, *supra* note 13, at 8. The  
18 same is true of the meaning of "discharge," which the Legislature determined was  
19 unnecessary to separately define.

20 F. The Water Board's Implied Authority to Regulate Does Not Extend to Air  
21 Emissions.

22 Obviously, if the Legislature had intended the water boards to have authority to  
23 investigate air emissions, it could have so stated in the statute. By confining itself to the  
24 same terminology used elsewhere in the statute, it must be concluded that the Legislature  
25 intended section 13267 to be limited to waterborne discharges. Thus, even if the  
26 information requested by the Water Board in the 13267 Requirement Letters would be  
27 helpful to determining or better understanding sources of pollution "affecting" water  
28 quality in the region, the Legislature has not authorized the Water Board to exert authority

1 to require private parties to develop this information. Nor can this authority be implied  
2 from the Water Board's broad mandate regarding "water quality control" and the necessary  
3 authority it carries to regulate activities which "may affect the quality of the waters of the  
4 State." Water Code, § 13000. As the court in *Addison* stated,

5 the doctrine of implied powers is not without limitations. It  
6 cannot be invoked where the grant of express powers  
7 clearly excludes the exercise of others, or where the  
8 claimed power is incompatible with, *or outside the scope*  
9 *of, the express power. For a power to be justified under*  
10 *the doctrine, it must be essential to the declared objects*  
*and purposes of the enabling act -- not simply*  
*convenient, but indispensable. Any reasonable doubt*  
*concerning the existence of the power is to be resolved*  
*against the agency.*

11 *Addison v. Dept. of Motor Vehicles* 69 Cal. App. 3d 486, 498 (Cal. App. Ct. 1977)  
12 (emphasis added). While Water Board staff may have an interest in studying airborne  
13 emissions and their connection to water quality impairment, that investigation cannot be  
14 compelled under the guise that air emissions constitute "discharges" under the Water Code.  
15 Nor may this investigation be imposed on the refineries because it is convenient to do so.  
16 To the extent there is any doubt surrounding these issues, it must be resolved against the  
17 agency.

18 Similar limitations exist on administrative agencies' subpoena power, a type of  
19 investigatory authority not unlike that exercised here. Subpoenas issued by administrative  
20 agencies will not be enforced where the investigation is outside the authority of the agency,  
21 the demand is too indefinite, or the information sought is not reasonably relevant to the  
22 investigation. *Peters v. United States*, 853 F.2d 692, 699 (9th Cir. 1988). Such  
23 requirements are based on the fact that the "authority of an administrative agency to issue  
24 subpoenas for investigatory purposes is created solely by statute." *Id.* At 696; accord  
25 *United States v. LaSalle Nat'l Bank*, 437 U.S. 298, 307 (1978). Indeed, counsel to the State  
26 Board has recognized that although the State and regional water boards have broad  
27 authority to investigate water quality, such authority is limited to actions "*authorized or*  
28 *required under Porter-Cologne.*" State Water Resources Control Board Chief Counsel's

1 Statement for California's Nonpoint Source Pollution Control Program, Appendix B: Legal  
2 Opinions (citing Water Code, § 13267(a)); *see also Wildlife Alive v. Chickering*, 553 P.2d  
3 537, 540 (Cal. 1976) (reasoning that "in the grants (of powers) and in the regulation of the  
4 mode of exercise, there is an implied negative; an implication that no other than the  
5 expressly granted power passes by the grant; that it is to be exercised only in the prescribed  
6 mode").

7 More importantly, California has vested authority over air emissions with local and  
8 regional air pollution control and air quality management districts located throughout the  
9 state (in Los Angeles, the SCAQMD). These air districts are "primarily responsible for  
10 control of air pollution from *all* sources, other than emission from motor vehicles." Health  
11 & Saf. Code, §§ 39002, 40000 (emphasis added). Where a statute includes a grant of  
12 authority to one agency, it is presumed that the grant of authority extends only to that  
13 agency. Thus, the granting of exclusive authority implies the negative: that other agencies  
14 do not have such authority and any action taken by such other agencies cannot conflict or  
15 interfere with such authority. *See State Highway Com. v. Rawson*, 312 P.2d 849, 864 (Or.  
16 1957) (rejecting construction of statute that would allow for interference and obstruction of  
17 one state agency by another in the latter's exercise of its statutory powers and duties). By  
18 placing authority to regulate stationary source emissions exclusively within the province of  
19 the air districts, the Legislature withheld such authority from the water boards, even in  
20 those instances where such emissions could be shown to adversely affect water quality.

21 In conclusion, no court has ever addressed whether air emissions are "discharges"  
22 under the Porter-Cologne Act and common sense dictates otherwise. If and when this  
23 question is ever presented, Petitioners submit the courts will be guided by the principles  
24 and precedents set forth herein and will not expand the Water Code in this ill-considered  
25 manner. If the Water Board wishes to compel an investigation (or regulation) of aerial  
26 deposition as a source of water pollution, it must go back to the Legislature and seek to  
27 obtain this authority. The legislative process can then be used — as it should be — to vet  
28 the advantages and disadvantages of this approach and to evaluate the myriad of interests

1 and considerations that bear upon the ultimate policy decision of whether to expand the  
2 water boards' authority in so significant a manner.

3 III. THE TMDL PROGRAM DOES NOT PROVIDE THE WATER BOARD  
4 AUTHORITY TO ORDER PRIVATE PARTIES TO INVESTIGATE  
5 SOURCES OF AIR EMISSIONS UNDER SECTION 13267.

6 A TMDL is essentially a pollutant "budget" which (i) determines the total amount  
7 of a pollutant that the water body can accept while still achieving the applicable water  
8 quality standard, and (ii) allocates that total amount among all sources of the pollutant,  
9 including "point sources" (e.g., industrial and municipal wastewater discharges) and "non-  
10 point sources" (e.g., storm water runoff and agriculture). See 40 C.F.R. § 130.7.<sup>16</sup>  
11 Petitioners acknowledge that EPA considers sources of atmospheric deposition to be  
12 nonpoint sources under the TMDL program. See *Draft Guidance for Water Quality Based*  
13 *Decisions: The TMDL Process (Second Edition)* (EPA August 1999, EPA-84-D-99-0001)  
14 ("Draft EPA TMDL Guidance"), at Chapter 3.1.<sup>17</sup>

15 Nevertheless, "a TMDL does not, by itself, prohibit any conduct or require any  
16 actions. Instead, each TMDL represents a goal that may be implemented by adjusting  
17 pollutant discharge requirements in individual NPDES permits or establishing nonpoint  
18 source controls." *City of Arcadia v. EPA*, 265 F.Supp.2d 1142, 1145 (N.D. Cal 2003).  
19 "Thus, a TMDL forms the basis for *further* administrative actions that may require or  
20 prohibit conduct with respect to particularized pollutant discharges and waterbodies." *Id.*  
21 (emphasis added). In other words, the TMDL process, in and of itself, does not imbue the  
22 State or regional water boards with any new authority. Rather, TMDLs must be developed  
23 and implemented through existing regulatory programs (or through newly developed  
24 programs if authorized by law and duly adopted in accordance with the Administrative

25 <sup>16</sup> TMDLs can be established for waterbodies impacted solely from non-point sources.  
26 See *Pronsolino v. Nastri*, 291 F.3d 1123, 1139-41 (9th Cir. 2002).

27 <sup>17</sup> Interestingly, in 2002, the definition of "load allocation" (i.e., the portion of the total  
28 load allocated to a non-point source) included "atmospheric deposition" but that phrase was  
deleted from the definition in 2003.

1 Procedures Act). As EPA guidance points out, the TMDL implementation plan “must  
2 contain a *description of the legal authorities under which implementation will occur*. These  
3 authorities include, but are not limited to, NPDES, § 401 certification, Federal Land Policy  
4 Management Act § 202, the [Coastal Zone Act Reauthorization Amendments], State forest  
5 practices acts, CWA § 319 management programs, and various State, Territorial, Tribal and  
6 local programs.” *Draft EPA TMDL Guidance*, at Chapter 3.1 (emphasis added). The  
7 reduction goals (or allocations) in a TMDL are enforceable only to the extent legal  
8 authority exists to achieve the reductions. Insofar as air emissions are concerned, this  
9 authority does not reside in Water Code section 13267 or any other provision of the Water  
10 Code.<sup>18</sup>

11 IV. PRACTICAL CONSIDERATIONS MILITATE STRONGLY AGAINST  
12 ALLOWING THE WATER BOARDS TO INVESTIGATE AIR  
13 EMISSIONS ON AN AD HOC BASIS.

14 EPA has long recognized the technical difficulties associated with implementing a  
15 program for regulating atmosphere deposition as part of a TMDL program. *See* 64 Fed.  
16 Reg. at 46,022-23 (August 23, 1999) (“EPA recognizes that data, analytical approaches and  
17 models to establish TMDLs for pollutants originating from air deposition may not be  
18 immediately available, especially for pollutants subject to long range transport in the  
19 atmosphere”); *see also Technical Guidance Manual for Developing Total Maximum Daily*  
20

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21 <sup>18</sup> Several TMDLs recently approved by EPA include load allocations for nonpoint  
22 sources of atmospheric deposition. *See, e.g., Minnesota Statewide Total Maximum Daily*  
23 *Load, Minnesota Pollution Control Agency* (approved by EPA, March 27, 2007)  
24 (“*Minnesota TMDL*”); *Total Maximum Daily Load (TMDL) For Total Mercury in Fish*  
25 *Tissue Residue In the Middle & Lower Savannah River Watershed* U.S. EPA, Region IV  
26 (February 28, 2001) (“*Savannah River TMDL*”) However, implementation of plans to  
27 reduce emissions causing air deposition did not come from regulatory programs  
28 administered by state water agencies. Rather, these TMDLs proposed that reductions in air  
emissions be accomplished pursuant to state and federal *air* regulatory programs. *See, e.g.,*  
*Savannah River TMDL*, at 9 (“EPA expects that emissions of mercury from air sources (and  
consequently deposition of mercury to the Savannah River) will continue to be reduced  
during [] this TMDL through implementation of the CAA’s [] regulations. . . . EPA is  
considering additional regulatory actions under the CAA that may result in further  
reductions of mercury emissions from air sources.”).

1 *Loads, Book II: Part 1*, at 2-3. ("The control of atmospheric deposition is usually addressed  
2 in regional and national programs and should be considered as part of the uncontrollable  
3 load for typical TMDL development").

4 The 13267 Requirement Letters, and the NRDC petition which triggered their  
5 issuance, selectively focus on a tiny fraction of local sources that potentially may contribute  
6 to direct and indirect air deposition to surface waters in the region. The Los Angeles Basin  
7 contains over four million residents and their motor vehicles, as well as many thousands of  
8 small and large business operations. One recent study demonstrates that dust from paved  
9 roads, containing contaminants from vehicle exhaust and vehicle brake and tire wear,  
10 represents the largest source of particle-associated metals emitted to the atmosphere and  
11 locally deposited. *See* Sabin et al. (2006), attached as Exhibit 6.<sup>19</sup> Moreover, air deposition  
12 is not merely a local phenomenon. On the contrary, pollutants deposited to surface waters  
13 in California have been traced to distant sources in Asia. *See, e.g.,* Steding and Flagle  
14 (2002), abstract attached as Exhibit 7.

15 Taken together, the myriad local and global sources of air emissions contribute far  
16 more to air deposition than the sources which NRDC, followed by the Water Board, chose  
17 to single out. NRDC's letter identified these facilities on the basis of Toxic Release  
18 Inventory (TRI) data reported to EPA. All that the TRI data indicate is that those individual  
19 stationary sources reported greater emissions of the specified chemicals as compared to  
20 other stationary sources. This result does not, by itself, convert stationary sources into a  
21 major source of air deposition.

22 The 13267 Requirement Letter states no rational basis to assume that its requirement  
23 for air deposition studies by a few facilities will provide any useful information for  
24 addressing the actual effects of air deposition on water quality. A proper study would need

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25 <sup>19</sup> NRDC cites an earlier study by Sabin et al. to imply that air emissions from the  
26 facilities it identified are a substantial source of deposition to water bodies in Los Angeles.  
27 *See* NRDC letter, page 6, n.33. On the contrary, the earlier study cited by NRDC did not  
28 investigate the sources of air deposition, while the Sabin et al. study attached as Exhibit 6  
identifies road dust as the largest source.

1 to address all significant sources of deposition and would require the joint efforts of a wide  
2 variety of agencies, municipalities and other entities. By contrast, focusing on a few  
3 facilities, on an ad hoc basis, is unlikely to yield any substantial benefit.

4  
5 **REQUEST FOR RELIEF**

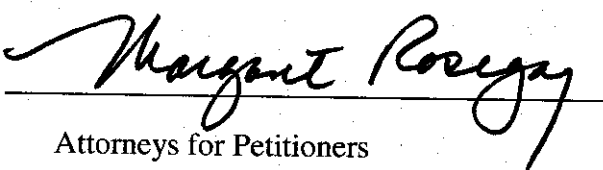
6 For the reasons set forth above, Petitioners respectfully request that the State Board  
7 grant Petitioners the following relief:

8 A. That the 13267 Requirement Letters issued to Refinery Petitioners be  
9 rescinded by the State Board.

10 B. Such other relief as the State Board may deem just and proper.

11  
12 Dated: June 14, 2007.

13  
14 PILLSBURY WINTHROP SHAW PITTMAN  
15 LLP  
16 MARGARET ROSEGAY  
17 NORMAN CARLIN  
18 50 Fremont Street  
19 Post Office Box 7880  
20 San Francisco, CA 94120-7880

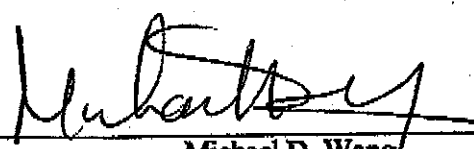
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VERIFICATION

I, Michael D. Wang, am Senior Advisor, Southern California, for the Western States Petroleum Association and have responsibility for oversight of water quality regulatory and policy matters at WSPA member facilities located in the Los Angeles Area. I have read the foregoing Verified Petition for Review and Request for Hearing and believe that the statements made therein are true and correct. If called as a witness to testify with respect to the matters stated therein, I could and would competently do so under oath.

I declare under penalty of perjury under the laws of the State of California that the foregoing is true and correct and that this verification was executed in Los Angeles, California, on June 14, 2007.

  
\_\_\_\_\_  
Michael D. Wang







# California Regional Water Quality Control Board

## Los Angeles Region



Linda S. Adams  
Cal/EPA Secretary

320 W. 4th Street, Suite 200, Los Angeles, California 90013  
Phone (213) 576-6600 FAX (213) 576-6640 - Internet Address: <http://www.waterboards.ca.gov/losangeles>

Arnold Schwarzenegger  
Governor

May 15, 2007

Shirley Tea  
Chevron Products Co.  
Chevron USA Inc  
324 W. El Segundo Blvd.  
El Segundo, CA 90245

### **REQUIREMENT UNDER CALIFORNIA WATER CODE SECTION 13267 FOR SUBMITTAL OF TECHNICAL REPORTS ON THE FATE AND TRANSPORT OF METALS EMITTED FROM THE CHEVRON PRODUCTS CO., EL SEGUNDO REFINERY**

Dear Ms. Tea:

The California Regional Water Quality Control Board, Los Angeles Region (Regional Board) is the public agency with primary responsibility for the protection of ground and surface water quality within major portions of Los Angeles and Ventura Counties. As part of our efforts to protect water quality, pursuant to California Water Code (CWC) Section 13267, the Regional Board is investigating the fate and transport of metals emitted from the Chevron Products Co., El Segundo Refinery. The Regional Board is interested in discharges of metals from atmospheric deposition to the Santa Monica Bay Watershed Management Area and other watersheds within the Los Angeles Region.

Based on information obtained from the US Environmental Protection Agency (EPA) Toxics Release Inventory database, the Chevron Products Co., El Segundo Refinery is a significant source of mercury and zinc emitted to the atmosphere. Some portion of these emissions then enters the waterbodies via direct deposition to a waterbody surface or deposition to the watershed and subsequent transportation to a waterbody via urban runoff.

CWC §13267 provides the Regional Board with the authority to require the Chevron Products Co., El Segundo Refinery to furnish, under penalty of perjury, technical reports that the Regional Board requires. The El Segundo Refinery currently discharges waste under NPDES Permit No. CA0000337, Order No. R4-2006-0089. Pursuant to CWC §13267, you are hereby required to submit a technical report that evaluates the fate and transport of atmospheric emissions of metals as they relate to either direct or indirect deposition to the Santa Monica Bay Watershed Management Area. The report shall be based on a sampling and analysis program that accounts for intra- and inter-annual variability in emissions.

*California Environmental Protection Agency*



*Our mission is to preserve and enhance the quality of California's water resources for the benefit of present and future generations.*

The desired outcomes of the report are:

- An estimate of the total mass of mercury and zinc emitted directly to the atmosphere per year from the El Segundo Refinery.
- A thorough discussion of the estimation methodology, uncertainties in the estimate, and assumptions used in the calculation of the total mass emitted.
- A discussion of the fate of these emitted metals and an estimate of how much of these metals are discharged to the Santa Monica Bay via direct or indirect deposition<sup>1</sup>. Along with a discussion of the basis for these estimates, include a thorough discussion of estimation methodology, uncertainties in the estimates, and assumptions used in the calculations of the fate and transport of these metals. If it is determined that metals are deposited in other watersheds in the Los Angeles Region, include these results as well.

#### **Need for Technical Report**

The waterbodies in the Santa Monica Bay Watershed Management Area are impaired by metals and are included on the Clean Water Act Section 303(d) list for metals in the water column and sediment. The Ballona Creek TMDL is effective as of January 11, 2006. The Marina del Rey TMDL is effective as of March 22, 2006.

The Regional Board needs the information in the required reports to assess the significance of atmospheric deposition as a source of metals discharged to the Santa Monica Bay Watershed Management Area. In addition, this information will help to adapt implementation actions for facilities to reduce, if necessary, metals loading as part of the Ballona Creek and Marina del Rey TMDLs. The contribution of direct and indirect atmospheric deposition of metals is a source of uncertainty that needs to be resolved for successful TMDL implementation.

Dry deposition of metals represents a major loading pathway in arid regions like Los Angeles, where wet deposition is limited due to limited precipitation.<sup>2</sup> In a study conducted by Sabin et al.<sup>2</sup> indirect dry deposition to the land surface within a watershed had the potential to be a large influence on the quality of storm water runoff. Particles that deposit on urban surfaces during dry weather can be easily mobilized by surface flows during storms due to the extensive impervious

<sup>1</sup> Direct atmospheric deposition is the deposition of pollutants directly to the surface of a waterbody. Indirect atmospheric deposition is the process by which pollutants deposited on the land surface may be washed off during storm events or by urban runoff and delivered to a waterbody.

<sup>2</sup> Sabin, L.D., K. Schiff, J.H. Lim and K.D. Stolzenbach. 2004. *Atmospheric dry deposition of trace metals in the Los Angeles coastal region*.

urban areas. Mean deposition of metals to the surface area of the Ballona Creek watershed, the largest subwatershed of the Santa Monica Bay Watershed Management Area, was calculated to be 13,000 kg/year for zinc.<sup>2</sup> In addition, Sabin<sup>2</sup> calculated the ratio of storm water runoff to indirect atmospheric deposition as 29% for zinc.<sup>2</sup>

Based on information obtained from the US Environmental Protection Agency (EPA) Toxics Release Inventory database, the Chevron Products Co., El Segundo Refinery released atmospheric emissions of 13 lbs of mercury and 1,069 lbs of zinc in 2004.<sup>3</sup> Research strongly suggests that a portion of these emissions is being discharged either directly or indirectly, into Los Angeles water bodies through atmospheric deposition. One study found that about 95% of emissions in Santa Monica Bay come from emission sources in Los Angeles County, and that 99% of lead found in Santa Monica Bay entered the Bay through atmospheric deposition.<sup>4</sup>

#### **Burden and Cost of Technical Report**

The Chevron Products Co. already calculates and keeps records of emissions of metals as part of their federal Toxics Release Inventory requirements and South Coast Air Quality Management District (AQMD) annual emissions reporting requirements. The cost of providing these estimates to the Regional Board is therefore minimal. The effort to calculate the fate of these emitted metals and to estimate how much of these metals would be discharged to the Santa Monica Bay Watershed Management Area may present an additional cost. Such an estimate may require air dispersion modeling and an estimation of the deposition of metals to land and water surfaces. Air deposition flux rates can be determined from literature values. EPA-approved dispersion model programs are available from [www.epa.gov/scram001](http://www.epa.gov/scram001). Assuming that one person can perform the necessary air dispersion modeling and deposition estimates in two months, then the time to perform the analysis is approximately 340 hours. Therefore, assuming a burdened hourly rate of \$100 per hour, the estimated cost for this analysis would be \$34,000.

Because the El Segundo Refinery already conducts air dispersion modeling as part of the preparation of health risk assessments under the requirements of the California Air Toxics "Hot Spots" Information and Assessment Act of 1987 (AB 2588), the additional cost of conducting air dispersion modeling is minimal. Therefore, the burden, including the cost, of the technical report bears a reasonable relationship to the need for and benefits to be obtained from the technical report.

<sup>3</sup> EPA Toxics Release Inventory. 2004. Web site  
<http://www.epa.gov/triexplorer/statefactsheet.htm>.

<sup>4</sup> Stolzenbach, K.D., R. Lu, C. Xiong, S. Friedlander, R. Turco, K. Schiff and L. Tiefenthaler. 2001. *Measuring and Modeling of Atmospheric Deposition and Santa Monica Bay and the Santa Monica Bay Watershed*.



May 15, 2007

Pursuant to CWC §13267, Chevron Products Co., El Segundo Refinery must submit the required technical report by **September 7, 2007**. Furthermore, pursuant to §13268 (b)(1) of the CWC failure to submit the required report may result in the imposition of civil liability penalties by the Regional Board of up to \$1,000.00 per day for each day the report is not received after the **September 7, 2007** due date. These civil liabilities may be assessed by the Regional Board for failure to comply, beginning with the date that the violations first occurred, and without further warning.

If you have any questions, please contact Mr. Sam Unger at (213) 576-6622 or alternatively, Ms. Rebecca Christmann at (213) 576-6757 regarding this matter.

Sincerely,



Deborah J. Smith  
Interim Executive Officer

cc: Mr. Michael Levy Esq., Office of Chief Counsel, State Water Resources Control Board  
Mr. Bruce Fujimoto, Division of Water Quality, State Water Resources Control Board  
Mr. Gerald Bowes, Division of Water Quality, State Water Resources Control Board  
Ms. Kathi Moore, Office of the Director, Water Division, U.S. EPA Region IX  
Mr. Eugene Bromley, CWA Standards and Permits, U.S. EPA Region IX  
Ms. Elaine Chang, Deputy Executive Officer for Planning, Rule Development, and Area Sources, South Coast Air Quality Management District  
Mr. David Beckman, Natural Resources Defense Council

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# California Regional Water Quality Control Board

## Los Angeles Region



Linda S. Adams  
Cal/EPA Secretary

320 W. 4th Street, Suite 200, Los Angeles, California 90013  
Phone (213) 576-6600 FAX (213) 576-6640 - Internet Address: <http://www.waterboards.ca.gov/losangeles>

Arnold Schwarzenegger  
Governor

May 15, 2007

Paul Langland  
ConocoPhillips,  
Los Angeles Refinery, Carson Plant  
1520 E. Sepulveda Blvd.  
Carson, CA 90745

### **REQUIREMENT UNDER CALIFORNIA WATER CODE SECTION 13267 FOR SUBMITTAL OF TECHNICAL REPORTS ON THE FATE AND TRANSPORT OF METALS EMITTED FROM CONOCOPHILLIPS, LOS ANGELES REFINERY, CARSON PLANT**

Dear Mr. Langland:

The California Regional Water Quality Control Board, Los Angeles Region (Regional Board) is the public agency with primary responsibility for the protection of ground and surface water quality within major portions of Los Angeles and Ventura Counties. As part of our efforts to protect water quality, pursuant to California Water Code (CWC) Section 13267, the Regional Board is investigating the fate and transport of metals emitted from the ConocoPhillips, Los Angeles Refinery, Carson Plant. The Regional Board is interested in discharges of metals from atmospheric deposition to the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area and other watersheds within the Los Angeles Region.

Based on information obtained from the US Environmental Protection Agency (EPA) Toxics Release Inventory database, the ConocoPhillips, Los Angeles Refinery, Carson Plant is a significant source of mercury emitted to the atmosphere. Some portion of these emissions then enters the waterbodies via direct deposition to a waterbody surface or deposition to the watershed and subsequent transportation to a waterbody via urban runoff.

CWC §13267 provides the Regional Board with the authority to require the ConocoPhillips, Los Angeles Refinery, Carson Plant to furnish, under penalty of perjury, technical reports that the Regional Board requires. The Carson Plant currently discharges waste under NPDES Permit No. CA0063185, Order No. R4-2006-0082. Pursuant to CWC §13267, you are hereby required to submit a technical report that evaluates the fate and transport of atmospheric emissions of metals as they relate to either direct or indirect deposition to the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area. The report shall be based on a sampling and

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analysis program that accounts for intra- and inter-annual variability in emissions. The desired outcomes of the report are:

- An estimate of the total mass of mercury emitted directly to the atmosphere per year from the Carson Plant.
- A thorough discussion of the estimation methodology, uncertainties in the estimate, and assumptions used in the calculation of the total mass emitted.
- A discussion of the fate of these emitted metals and an estimate of how much of these metals are discharged to the Dominguez Channel, and Los Angeles and Long Beach Harbors via direct or indirect deposition<sup>1</sup>. Along with a discussion of the basis for these estimates, include a thorough discussion of estimation methodology, uncertainties in the estimates, and assumptions used in the calculations of the fate and transport of these metals. If it is determined that metals are deposited in other watersheds in the Los Angeles Region, include these results as well.

#### **Need for Technical Report**

The waterbodies in the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area are impaired by metals and are included on the 2006 Clean Water Act Section 303(d) list for metals in the water column, tissue, and sediment as well as for benthic community effects and sediment toxicity. The Regional Board is developing Total Maximum Daily Loads (TMDLs) to address these impairments. The Dominguez Channel, Los Angeles and Long Beach Harbors TMDL is scheduled for Regional Board consideration in 2008.

The Regional Board needs the information in the required reports to assess the significance of atmospheric deposition as a source of metals discharged to the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area. In addition, this information will help to adapt implementation actions for facilities to reduce, if necessary, metals loading as part of the upcoming Dominguez Channel, Los Angeles and Long Beach Harbors TMDLs. The contribution of direct and indirect atmospheric deposition of metals is a source of uncertainty that needs to be resolved for successful TMDL development and implementation.

Dry deposition of metals represents a major loading pathway in arid regions like Los Angeles, where wet deposition is limited due to limited precipitation.<sup>2</sup> In a study conducted by Sabin et

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<sup>1</sup> Direct atmospheric deposition is the deposition of pollutants directly to the surface of a waterbody. Indirect atmospheric deposition is the process by which pollutants deposited on the land surface may be washed off during storm events or by urban runoff and delivered to a waterbody.

<sup>2</sup> Sabin, L.D., K. Schiff, J.H. Lim and K.D. Stolzenbach. 2004. *Atmospheric dry deposition of trace metals in the Los Angeles coastal region*.



al.<sup>2</sup> indirect dry deposition to the land surface within a watershed had the potential to be a large influence on the quality of storm water runoff. Particles that deposit on urban surfaces during dry weather can be easily mobilized by surface flows during storms due to the extensive impervious urban areas. Mean deposition of metals to the surface area of the Dominguez Channel watershed was calculated to be 2,100 kilograms per year (kg/year) for copper, 1,600 kg/year for lead, and 9,400 kg/year for zinc.<sup>2</sup> In addition, Sabin<sup>2</sup> calculated the ratio of storm water runoff to indirect atmospheric deposition as 31% for copper, 14% for lead, and 43% for zinc.<sup>2</sup>

Based on information obtained from the US Environmental Protection Agency (EPA) Toxics Release Inventory database, the ConocoPhillips, Los Angeles Refinery, Carson Plant released atmospheric emissions of 2 pounds (lbs) of mercury.<sup>3</sup> Research strongly suggests that a portion of these emissions is being discharged either directly or indirectly, into Los Angeles water bodies through atmospheric deposition. One study found that about 95% of emissions in Santa Monica Bay come from emission sources in Los Angeles County, and that 99% of lead found in Santa Monica Bay entered the Bay through atmospheric deposition.<sup>4</sup>

#### **Burden and Cost of Technical Report**

The ConocoPhillips Carson Plant already calculates and keeps records of emissions of metals as part of their federal Toxics Release Inventory requirements and South Coast Air Quality Management District (AQMD) annual emissions reporting requirements. The cost of providing these estimates to the Regional Board is therefore minimal. The effort to calculate the fate of these emitted metals and to estimate how much of these metals would be discharged to the Dominguez Channel, Los Angeles and Long Beach Harbors may present an additional cost. Such an estimate may require air dispersion modeling and an estimation of the deposition of metals to land and water surfaces. Air deposition flux rates can be determined from literature values. EPA-approved dispersion model programs are available from [www.epa.gov/scram001](http://www.epa.gov/scram001). Assuming that one person can perform the necessary air dispersion modeling and deposition estimates in two months, then the time to perform the analysis is approximately 340 hours. Therefore, assuming a burdened hourly rate of \$100 per hour, the estimated cost for this analysis would be \$34,000.

Because the Carson Plant already conducts air dispersion modeling as part of the preparation of health risk assessments under the requirements of the California Air Toxics "Hot Spots" Information and Assessment Act of 1987 (AB 2588), the additional cost of conducting air

<sup>3</sup> EPA Toxics Release Inventory. 2004. Web site  
<http://www.epa.gov/triexplorer/statefactsheet.htm>.

<sup>4</sup> Stolzenbach, K.D., R. Lu, C. Xiong, S. Friedlander, R. Turco, K. Schiff and L. Tiefenthaler. 2001. *Measuring and Modeling of Atmospheric Deposition and Santa Monica Bay and the Santa Monica Bay Watershed*.

Mr. Paul Langland

- 4 -

May 15, 2007

dispersion modeling is minimal. Therefore, the burden, including the cost, of the technical report bears a reasonable relationship to the need for and benefits to be obtained from the technical report.

Pursuant to CWC §13267, the ConocoPhillips, Los Angeles Refinery, Carson Plant must submit the required technical report by **September 7, 2007**. Furthermore, pursuant to §13268 (b)(1) of the CWC failure to submit the required report may result in the imposition of civil liability penalties by the Regional Board of up to \$1,000.00 per day for each day the report is not received after the **September 7, 2007** due date. These civil liabilities may be assessed by the Regional Board for failure to comply, beginning with the date that the violations first occurred, and without further warning.

If you have any questions, please contact Mr. Sam Unger at (213) 576-6622 or alternatively, Ms. Rebecca Christmann at (213) 576-6757 regarding this matter.

Sincerely,



Deborah J. Smith  
Interim Executive Officer

cc: Mr. Michael Levy Esq., Office of Chief Counsel, State Water Resources Control Board  
Mr. Bruce Fujimoto, Division of Water Quality, State Water Resources Control Board  
Mr. Gerald Bowes, Division of Water Quality, State Water Resources Control Board  
Ms. Kathi Moore, Office of the Director, Water Division, U.S. EPA Region IX  
Mr. Eugene Bromley, CWA Standards and Permits, U.S. EPA Region IX  
Ms. Elaine Chang, Deputy Executive Officer for Planning, Rule Development, and Area Sources, South Coast Air Quality Management District  
Mr. David Beckman, Natural Resources Defense Council

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# California Regional Water Quality Control Board

## Los Angeles Region



Linda S. Adams  
Cal/EPA Secretary

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Phone (213) 576-6600 FAX (213) 576-6640 - Internet Address: <http://www.waterboards.ca.gov/losangeles>

Arnold Schwarzenegger  
Governor

May 15, 2007

Paul Langland  
ConocoPhillips  
Los Angeles Marine Terminal  
150 Pier A Street  
Wilmington, CA 90744

### REQUIREMENT UNDER CALIFORNIA WATER CODE SECTION 13267 FOR SUBMITTAL OF TECHNICAL REPORTS ON THE FATE AND TRANSPORT OF METALS EMITTED FROM THE CONOCOPHILLIPS, LOS ANGELES MARINE TERMINAL

Dear Mr. Langland:

The California Regional Water Quality Control Board, Los Angeles Region (Regional Board) is the public agency with primary responsibility for the protection of ground and surface water quality within major portions of Los Angeles and Ventura Counties. As part of our efforts to protect water quality, pursuant to California Water Code (CWC) Section 13267, the Regional Board is investigating the fate and transport of metals emitted from the ConocoPhillips, Los Angeles Marine Terminal. The Regional Board is interested in discharges of metals from atmospheric deposition to the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area and other watersheds within the Los Angeles Region.

Based on information obtained from the US Environmental Protection Agency (EPA) Toxics Release Inventory database, the ConocoPhillips, Los Angeles Marine Terminal is a significant source of mercury and zinc emitted to the atmosphere. Some portion of these emissions then enters the waterbodies via direct deposition to a waterbody surface or deposition to the watershed and subsequent transportation to a waterbody via urban runoff.

CWC §13267 provides the Regional Board with the authority to require the ConocoPhillips, Los Angeles Marine Terminal to furnish, under penalty of perjury, technical reports that the Regional Board requires. Pursuant to CWC §13267, you are hereby required to submit a technical report that evaluates the fate and transport of atmospheric emissions of metals as they relate to either direct or indirect deposition to the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area. The report shall be based on a sampling and

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analysis program that accounts for intra- and inter-annual variability in emissions. The desired outcomes of the report are:

- An estimate of the total mass of mercury and zinc emitted directly to the atmosphere per year from the Los Angeles Marine Terminal.
- A thorough discussion of the estimation methodology, uncertainties in the estimate, and assumptions used in the calculation of the total mass emitted.
- A discussion of the fate of these emitted metals and an estimate of how much of these metals are discharged to the Dominguez Channel, and Los Angeles and Long Beach Harbors via direct or indirect deposition<sup>1</sup>. Along with a discussion of the basis for these estimates, include a thorough discussion of estimation methodology, uncertainties in the estimates, and assumptions used in the calculations of the fate and transport of these metals. If it is determined that metals are deposited in other watersheds in the Los Angeles Region, include these results as well.

#### **Need for Technical Report**

The waterbodies in the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area are impaired by metals and are included on the 2006 Clean Water Act Section 303(d) list for metals in the water column, tissue, and sediment as well as for benthic community effects and sediment toxicity. The Regional Board is developing Total Maximum Daily Loads (TMDLs) to address these impairments. The Dominguez Channel, Los Angeles and Long Beach Harbors TMDL is scheduled for Regional Board consideration in 2008.

The Regional Board needs the information in the required reports to assess the significance of atmospheric deposition as a source of metals discharged to the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area. In addition, this information will help to adapt implementation actions for facilities to reduce, if necessary, metals loading as part of the upcoming Dominguez Channel, Los Angeles and Long Beach Harbors TMDLs. The contribution of direct and indirect atmospheric deposition of metals is a source of uncertainty that needs to be resolved for successful TMDL development and implementation.

Dry deposition of metals represents a major loading pathway in arid regions like Los Angeles, where wet deposition is limited due to limited precipitation.<sup>2</sup> In a study conducted by Sabin et

<sup>1</sup> Direct atmospheric deposition is the deposition of pollutants directly to the surface of a waterbody. Indirect atmospheric deposition is the process by which pollutants deposited on the land surface may be washed off during storm events or by urban runoff and delivered to a waterbody.

<sup>2</sup> Sabin, L.D., K. Schiff, J.H. Lim and K.D. Stolzenbach. 2004. *Atmospheric dry deposition of trace metals in the Los Angeles coastal region*.

al.<sup>2</sup> indirect dry deposition to the land surface within a watershed had the potential to be a large influence on the quality of storm water runoff. Particles that deposit on urban surfaces during dry weather can be easily mobilized by surface flows during storms due to the extensive impervious urban areas. Mean deposition of metals to the surface area of the Dominguez Channel watershed was calculated to be 2,100 kilograms per year (kg/year) for copper, 1,600 kg/year for lead, and 9,400 kg/year for zinc.<sup>2</sup> In addition, Sabin<sup>2</sup> calculated the ratio of storm water runoff to indirect atmospheric deposition as 31% for copper, 14% for lead, and 43% for zinc.<sup>2</sup>

Based on information obtained from the US Environmental Protection Agency (EPA) Toxics Release Inventory database, the ConocoPhillips, Los Angeles Marine Terminal released atmospheric emissions of 4 pounds (lbs) of mercury and 750 lbs of zinc in 2004.<sup>3</sup> Research strongly suggests that a portion of these emissions is being discharged either directly or indirectly, into Los Angeles water bodies through atmospheric deposition. One study found that about 95% of emissions in Santa Monica Bay come from emission sources in Los Angeles County, and that 99% of lead found in Santa Monica Bay entered the Bay through atmospheric deposition.<sup>4</sup>

#### **Burden and Cost of Technical Report**

ConocoPhillips, Los Angeles Marine Terminal already calculates and keeps records of emissions of metals as part of their federal Toxics Release Inventory requirements and South Coast Air Quality Management District (AQMD) annual emissions reporting requirements. The cost of providing these estimates to the Regional Board is therefore minimal. The effort to calculate the fate of these emitted metals and to estimate how much of these metals would be discharged to the Dominguez Channel, Los Angeles and Long Beach Harbors may present an additional cost. Such an estimate may require air dispersion modeling and an estimation of the deposition of metals to land and water surfaces. Air deposition flux rates can be determined from literature values. EPA-approved dispersion model programs are available from [www.epa.gov/scram001](http://www.epa.gov/scram001). Assuming that one person can perform the necessary air dispersion modeling and deposition estimates in two months, then the time to perform the analysis is approximately 340 hours. Therefore, assuming a burdened hourly rate of \$100 per hour, the estimated cost for this analysis would be \$34,000.

Because the Los Angeles Marine Terminal already conducts air dispersion modeling as part of the preparation of health risk assessments under the requirements of the California Air Toxics

<sup>3</sup> EPA Toxics Release Inventory. 2004. Web site  
<http://www.epa.gov/triexplorer/statefactsheet.htm>.

<sup>4</sup> Stolzenbach, K.D., R. Lu, C. Xiong, S. Friedlander, R. Turco, K. Schiff and L. Tiefenthaler. 2001. *Measuring and Modeling of Atmospheric Deposition and Santa Monica Bay and the Santa Monica Bay Watershed*.



Mr. Paul Langland

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"Hot Spots" Information and Assessment Act of 1987 (AB 2588), the additional cost of conducting air dispersion modeling is minimal. Therefore, the burden, including the cost, of the technical report bears a reasonable relationship to the need for and benefits to be obtained from the technical report.

Pursuant to CWC §13267, ConocoPhillips, Los Angeles Marine Terminal must submit the required technical report by **September 7, 2007**. Furthermore, pursuant to §13268 (b)(1) of the CWC failure to submit the required report may result in the imposition of civil liability penalties by the Regional Board of up to \$1,000.00 per day for each day the report is not received after the **September 7, 2007** due date. These civil liabilities may be assessed by the Regional Board for failure to comply, beginning with the date that the violations first occurred, and without further warning.

If you have any questions, please contact Mr. Sam Unger at (213) 576-6622 or alternatively, Ms. Rebecca Christmann at (213) 576-6757 regarding this matter.

Sincerely,



Deborah J. Smith  
Interim Executive Officer

cc: Mr. Michael Levy Esq., Office of Chief Counsel, State Water Resources Control Board  
Mr. Bruce Fujimoto, Division of Water Quality, State Water Resources Control Board  
Mr. Gerald Bowes, Division of Water Quality, State Water Resources Control Board  
Ms. Kathi Moore, Office of the Director, Water Division, U.S. EPA Region IX  
Mr. Eugene Bromley, CWA Standards and Permits, U.S. EPA Region IX  
Ms. Elaine Chang, Deputy Executive Officer for Planning, Rule Development, and Area Sources, South Coast Air Quality Management District  
Mr. David Beckman, Natural Resources Defense Council

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## California Regional Water Quality Control Board Los Angeles Region



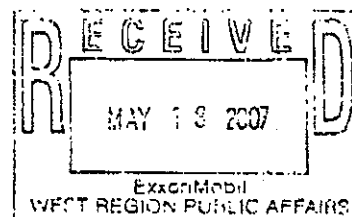
Linda S. Adams  
Cal/EPA Secretary

320 W. 4th Street, Suite 200, Los Angeles, California 90013  
Phone (213) 576-6600 FAX (213) 576-6640 - Internet Address: <http://www.waterboards.ca.gov/losangeles>

Arnold Schwarzenegger  
Governor

May 15, 2007

Carolyn A. Keith  
ExxonMobil Oil Corporation  
Torrance Refinery  
3700 West 190<sup>th</sup> St.  
Torrance, CA 90509



### REQUIREMENT UNDER CALIFORNIA WATER CODE SECTION 13267 FOR SUBMITTAL OF TECHNICAL REPORTS ON THE FATE AND TRANSPORT OF METALS EMITTED FROM THE EXXONMOBIL OIL CORPORATION, TORRANCE REFINERY

Dear Ms. Keith:

The California Regional Water Quality Control Board, Los Angeles Region (Regional Board) is the public agency with primary responsibility for the protection of ground and surface water quality within major portions of Los Angeles and Ventura Counties. As part of our efforts to protect water quality, pursuant to California Water Code (CWC) Section 13267, the Regional Board is investigating the fate and transport of metals emitted from the ExxonMobil Oil Corporation, Torrance Refinery. The Regional Board is interested in discharges of metals from atmospheric deposition to the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area and other watersheds within the Los Angeles Region.

Based on information obtained from the US Environmental Protection Agency (EPA) Toxics Release Inventory database, the ExxonMobil Corporation, Torrance Refinery is a significant source of copper, lead, mercury, and zinc emitted to the atmosphere. Some portion of these emissions then enters the waterbodies via direct deposition to a waterbody surface or deposition to the watershed and subsequent transportation to a waterbody via urban runoff.

CWC §13267 provides the Regional Board with the authority to require the ExxonMobil Corporation, Torrance Refinery to furnish, under penalty of perjury, technical reports that the Regional Board requires. The Torrance Refinery currently discharges waste under NPDES Permit No. CA0055387, Order No. R4-2001-129. Pursuant to CWC §13267, you are hereby required to submit a technical report that evaluates the fate and transport of atmospheric emissions of metals as they relate to either direct or indirect deposition to the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area. The report shall

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Ms. Carolin A. Keith

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May 15, 2007

be based on a sampling and analysis program that accounts for intra- and inter-annual variability in emissions. The desired outcomes of the report are:

- An estimate of the total mass of copper, lead, mercury, and zinc emitted directly to the atmosphere per year from the Torrance Refinery.
- A thorough discussion of the estimation methodology, uncertainties in the estimate, and assumptions used in the calculation of the total mass emitted.
- A discussion of the fate of these emitted metals and an estimate of how much of these metals are discharged to the Dominguez Channel, and Los Angeles and Long Beach Harbors via direct or indirect deposition<sup>1</sup>. Along with a discussion of the basis for these estimates, include a thorough discussion of estimation methodology, uncertainties in the estimates, and assumptions used in the calculations of the fate and transport of these metals. If it is determined that metals are deposited in other watersheds in the Los Angeles Region, include these results as well.

#### Need for Technical Report

The waterbodies in the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area are impaired by metals and are included on the 2006 Clean Water Act Section 303(d) list for metals in the water column, tissue, and sediment as well as for benthic community effects and sediment toxicity. The Regional Board is developing Total Maximum Daily Loads (TMDLs) to address these impairments. The Dominguez Channel, Los Angeles and Long Beach Harbors TMDL is scheduled for Regional Board consideration in 2008.

The Regional Board needs the information in the required reports to assess the significance of atmospheric deposition as a source of metals discharged to the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area. In addition, this information will help to adapt implementation actions for facilities to reduce, if necessary, metals loading as part of the upcoming Dominguez Channel, Los Angeles and Long Beach Harbors TMDLs. The contribution of direct and indirect atmospheric deposition of metals is a source of uncertainty that needs to be resolved for successful TMDL development and implementation.

Dry deposition of metals represents a major loading pathway in arid regions like Los Angeles, where wet deposition is limited due to limited precipitation.<sup>2</sup> In a study conducted by Sabin et

<sup>1</sup> Direct atmospheric deposition is the deposition of pollutants directly to the surface of a waterbody. Indirect atmospheric deposition is the process by which pollutants deposited on the land surface may be washed off during storm events or by urban runoff and delivered to a waterbody.

<sup>2</sup> Sabin, L.D., K. Schiff, J.H. Lim and K.D. Stolzenbach. 2004. *Atmospheric dry deposition of trace metals in the Los Angeles coastal region.*

Ms. Carolin A. Keith

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May 15, 2007

al.<sup>2</sup> indirect dry deposition to the land surface within a watershed had the potential to be a large influence on the quality of storm water runoff. Particles that deposit on urban surfaces during dry weather can be easily mobilized by surface flows during storms due to the extensive impervious urban areas. Mean deposition of metals to the surface area of the Dominguez Channel watershed was calculated to be 2,100 kilograms per year (kg/year) for copper, 1,600 kg/year for lead, and 9,400 kg/year for zinc.<sup>2</sup> In addition, Sabin<sup>2</sup> calculated the ratio of storm water runoff to indirect atmospheric deposition as 31% for copper, 14% for lead, and 43% for zinc.<sup>2</sup>

Based on information obtained from the US Environmental Protection Agency (EPA) Toxics Release Inventory database, the ExxonMobil Corporation, Torrance Refinery released atmospheric emissions of 836 pounds (lbs) of copper, 212 lbs of lead, 115 lbs of mercury, and 13,350 lbs of zinc in 2004.<sup>3</sup> Research strongly suggests that a portion of these emissions is being discharged either directly or indirectly, into Los Angeles water bodies through atmospheric deposition. One study found that about 95% of emissions in Santa Monica Bay come from emission sources in Los Angeles County, and that 99% of lead found in Santa Monica Bay entered the Bay through atmospheric deposition.<sup>4</sup>

#### **Burden and Cost of Technical Report**

The ExxonMobil Corporation already calculates and keeps records of emissions of metals as part of their federal Toxics Release Inventory requirements and South Coast Air Quality Management District (AQMD) annual emissions reporting requirements. The cost of providing these estimates to the Regional Board is therefore minimal. The effort to calculate the fate of these emitted metals and to estimate how much of these metals would be discharged to the Dominguez Channel, Los Angeles and Long Beach Harbors may present an additional cost. Such an estimate may require air dispersion modeling and an estimation of the deposition of metals to land and water surfaces. Air deposition flux rates can be determined from literature values. EPA-approved dispersion model programs are available from [www.epa.gov/scram001](http://www.epa.gov/scram001). Assuming that one person can perform the necessary air dispersion modeling and deposition estimates in two months, then the time to perform the analysis is approximately 340 hours. Therefore, assuming a burdened hourly rate of \$100 per hour, the estimated cost for this analysis would be \$34,000.

Because the Torrance Refinery already conducts air dispersion modeling as part of the preparation of health risk assessments under the requirements of the California Air Toxics "Hot Spots" Information and Assessment Act of 1987 (AB 2588), the additional cost of conducting air

<sup>3</sup> EPA Toxics Release Inventory. 2004. Web site  
<http://www.epa.gov/triexplorer/statefactsheet.htm>.

<sup>4</sup> Stolzenbach, K.D., R. Lu, C. Xiong, S. Friedlander, R. Turco, K. Schiff and L. Tiefenthaler. 2001. *Measuring and Modeling of Atmospheric Deposition and Santa Monica Bay and the Santa Monica Bay Watershed*.

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Ms. Carolin A. Keith

- 4 -

May 15, 2007

dispersion modeling is minimal. Therefore, the burden, including the cost, of the technical report bears a reasonable relationship to the need for and benefits to be obtained from the technical report.

Pursuant to CWC §13267, ExxonMobil Corporation, Torrance Refinery must submit the required technical report by September 7, 2007. Furthermore, pursuant to §13268 (b)(1) of the CWC failure to submit the required report may result in the imposition of civil liability penalties by the Regional Board of up to \$1,000.00 per day for each day the report is not received after the September 7, 2007 due date. These civil liabilities may be assessed by the Regional Board for failure to comply, beginning with the date that the violations first occurred, and without further warning.

If you have any questions, please contact Mr. Sam Unger at (213) 576-6622 or alternatively, Ms. Rebecca Christmann at (213) 576-6757 regarding this matter.

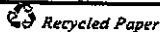
Sincerely,



Deborah J. Smith  
Interim Executive Officer

cc: Mr. Michael Levy Esq., Office of Chief Counsel, State Water Resources Control Board  
Mr. Bruce Fujimoto, Division of Water Quality, State Water Resources Control Board  
Mr. Gerald Bowes, Division of Water Quality, State Water Resources Control Board  
Ms. Kathi Moore, Office of the Director, Water Division, U.S. EPA Region IX  
Mr. Eugene Bromley, CWA Standards and Permits, U.S. EPA Region IX  
Ms. Elaine Chang, Deputy Executive Officer for Planning, Rule Development, and Area Sources, South Coast Air Quality Management District  
Mr. David Beckman, Natural Resources Defense Council

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# California Regional Water Quality Control Board

## Los Angeles Region



Linda S. Adams  
Cal/EPA Secretary

320 W. 4th Street, Suite 200, Los Angeles, California 90013  
Phone (213) 576-6600 FAX (213) 576-6640 - Internet Address: <http://www.waterboards.ca.gov/losangeles>

Arnold Schwarzenegger  
Governor

May 15, 2007

Mr. Walter W. Neil  
BP West Coast Products LLC, Carson  
1801 E. Sepulveda Blvd.  
Carson, CA 90749

### **REQUIREMENT UNDER CALIFORNIA WATER CODE SECTION 13267 FOR SUBMITTAL OF TECHNICAL REPORTS ON THE FATE AND TRANSPORT OF METALS EMITTED FROM BP WEST COAST PRODUCTS LLC, CARSON REFINERY**

Dear Mr. Neil:

The California Regional Water Quality Control Board, Los Angeles Region (Regional Board) is the public agency with primary responsibility for the protection of ground and surface water quality within major portions of Los Angeles and Ventura Counties. As part of our efforts to protect water quality, pursuant to California Water Code (CWC) Section 13267, the Regional Board is investigating the fate and transport of metals emitted from the BP Carson Refinery. The Regional Board is interested in discharges of metals from atmospheric deposition to the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area and other watersheds within the Los Angeles Region.

Based on information obtained from the US Environmental Protection Agency (EPA) Toxics Release Inventory database, the BP Carson Refinery is a significant source of lead, mercury, and zinc emitted to the atmosphere. Some portion of these emissions then enters the waterbodies via direct deposition to a waterbody surface or deposition to the watershed and subsequent transportation to a waterbody via urban runoff.

CWC §13267 provides the Regional Board with the authority to require the BP Carson Refinery to furnish, under penalty of perjury, technical reports that the Regional Board requires. BP Carson Refinery currently discharges waste under NPDES Permit No. CA0000680, Order No. R4-2007-0015. Pursuant to CWC §13267, you are hereby required to submit a technical report that evaluates the fate and transport of atmospheric emissions of metals as they relate to either direct or indirect deposition to the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area. The report shall be based on a sampling and analysis program that accounts for intra- and inter-annual variability in emissions.

*California Environmental Protection Agency*



*Our mission is to preserve and enhance the quality of California's water resources for the benefit of present and future generations.*

The desired outcomes of the report are:

- An estimate of the total mass of lead, mercury, and zinc emitted directly to the atmosphere per year from the BP Carson Refinery.
- A thorough discussion of the estimation methodology, uncertainties in the estimate, and assumptions used in the calculation of the total mass emitted.
- A discussion of the fate of these emitted metals and an estimate of how much of these metals are discharged to the Dominguez Channel, and Los Angeles and Long Beach Harbors via direct or indirect deposition<sup>1</sup>. Along with a discussion of the basis for these estimates, include a thorough discussion of estimation methodology, uncertainties in the estimates, and assumptions used in the calculations of the fate and transport of these metals. If it is determined that metals are deposited in other watersheds in the Los Angeles Region, include these results as well.

#### Need for Technical Report

The waterbodies in the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area are impaired by metals and are included on the 2006 Clean Water Act Section 303(d) list for metals in the water column, tissue, and sediment as well as for benthic community effects and sediment toxicity. The Regional Board is developing Total Maximum Daily Loads (TMDLs) to address these impairments. The Dominguez Channel, Los Angeles and Long Beach Harbors TMDL is scheduled for Regional Board consideration in 2008.

The Regional Board needs the information in the required reports to assess the significance of atmospheric deposition as a source of metals discharged to the Dominguez Channel, Los Angeles and Long Beach Harbors Watershed Management Area. In addition, this information will help to adapt implementation actions for facilities to reduce, if necessary, metals loading as part of the upcoming Dominguez Channel, Los Angeles and Long Beach Harbors TMDLs. The contribution of direct and indirect atmospheric deposition of metals is a source of uncertainty that needs to be resolved for successful TMDL development and implementation.

Dry deposition of metals represents a major loading pathway in arid regions like Los Angeles, where wet deposition is limited due to limited precipitation.<sup>2</sup> In a study conducted by Sabin et al.<sup>2</sup> indirect dry deposition to the land surface within a watershed had the potential to be a large

<sup>1</sup> Direct atmospheric deposition is the deposition of pollutants directly to the surface of a waterbody. Indirect atmospheric deposition is the process by which pollutants deposited on the land surface may be washed off during storm events or by urban runoff and delivered to a waterbody.

<sup>2</sup> Sabin, L.D., K. Schiff, J.H. Lim and K.D. Stolzenbach. 2004. *Atmospheric dry deposition of trace metals in the Los Angeles coastal region*.



influence on the quality of storm water runoff. Particles that deposit on urban surfaces during dry weather can be easily mobilized by surface flows during storms due to the extensive impervious urban areas. Mean deposition of metals to the surface area of the Dominguez Channel watershed was calculated to be 2,100 kilograms per year (kg/year) for copper, 1,600 kg/year for lead, and 9,400 kg/year for zinc.<sup>2</sup> In addition, Sabin<sup>2</sup> calculated the ratio of storm water runoff to indirect atmospheric deposition as 31% for copper, 14% for lead, and 43% for zinc.<sup>2</sup>

Based on information obtained from the US Environmental Protection Agency (EPA) Toxics Release Inventory database, the BP Carson Refinery released atmospheric emissions of 188 pounds (lbs) of lead, 166 lbs of mercury, and 1,330 lbs of zinc in 2004.<sup>3</sup> Research strongly suggests that a portion of these emissions is being discharged either directly or indirectly, into Los Angeles water bodies through atmospheric deposition. One study found that about 95% of emissions in Santa Monica Bay come from emission sources in Los Angeles County, and that 99% of lead found in Santa Monica Bay entered the Bay through atmospheric deposition.<sup>4</sup>

#### **Burden and Cost of Technical Report**

The BP Carson Refinery already calculates and keeps records of emissions of metals as part of their federal Toxics Release Inventory requirements and South Coast Air Quality Management District (AQMD) annual emissions reporting requirements. The cost of providing these estimates to the Regional Board is therefore minimal. The effort to calculate the fate of these emitted metals and to estimate how much of these metals would be discharged to the Dominguez Channel, Los Angeles and Long Beach Harbors may present an additional cost. Such an estimate may require air dispersion modeling and an estimation of the deposition of metals to land and water surfaces. Air deposition flux rates can be determined from literature values. EPA-approved dispersion model programs are available from [www.epa.gov/scram001](http://www.epa.gov/scram001). Assuming that one person can perform the necessary air dispersion modeling and deposition estimates in two months, then the time to perform the analysis is approximately 340 hours. Therefore, assuming a burdened hourly rate of \$100 per hour, the estimated cost for this analysis would be \$34,000.

Because the BP Carson Refinery already conducts air dispersion modeling as part of the preparation of health risk assessments under the requirements of the California Air Toxics "Hot Spots" Information and Assessment Act of 1987 (AB 2588), the additional cost of conducting air dispersion modeling is minimal. Therefore, the burden, including the cost, of the technical report

<sup>3</sup> EPA Toxics Release Inventory. 2004. Web site  
<http://www.epa.gov/triexplorer/statefactsheet.htm>.

<sup>4</sup> Stolzenbach, K.D., R. Lu, C. Xiong, S. Friedlander, R. Turco, K. Schiff and L. Tiefenthaler. 2001. *Measuring and Modeling of Atmospheric Deposition and Santa Monica Bay and the Santa Monica Bay Watershed*.





May 15, 2007

bears a reasonable relationship to the need for and benefits to be obtained from the technical report.

Pursuant to CWC §13267, the BP West Coast Products LLC, Carson Refinery must submit the required technical report by **September 7, 2007**. Furthermore, pursuant to §13268 (b)(1) of the CWC failure to submit the required report may result in the imposition of civil liability penalties by the Regional Board of up to \$1,000.00 per day for each day the report is not received after the **September 7, 2007** due date. These civil liabilities may be assessed by the Regional Board for failure to comply, beginning with the date that the violations first occurred, and without further warning.

If you have any questions, please contact Mr. Sam Unger at (213) 576-6622 or alternatively, Ms. Rebecca Christmann at (213) 576-6757 regarding this matter.

Sincerely,



Deborah J. Smith  
Interim Executive Officer

cc: Mr. Michael Levy Esq., Office of Chief Counsel, State Water Resources Control Board  
Mr. Bruce Fujimoto, Division of Water Quality, State Water Resources Control Board  
Mr. Gerald Bowes, Division of Water Quality, State Water Resources Control Board  
Ms. Kathi Moore, Office of the Director, Water Division, U.S. EPA Region IX  
Mr. Eugene Bromley, CWA Standards and Permits, U.S. EPA Region IX  
Ms. Elaine Chang, Deputy Executive Officer for Planning, Rule Development, and Area Sources, South Coast Air Quality Management District  
Mr. David Beckman, Natural Resources Defense Council

***California Environmental Protection Agency***



*Our mission is to preserve and enhance the quality of California's water resources for the benefit of present and future generations.*





## Dry deposition and resuspension of particle-associated metals near a freeway in Los Angeles

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### Abstract

Dry atmospheric deposition represents a potentially large source of pollutant metal contamination in urban stormwater runoff, yet there is a limited amount of research on the relationship between atmospheric emissions and water quality problems in urban areas. In Los Angeles, with air quality that ranks among the worst in the United States, significant quantities of toxic materials are released into the atmosphere every day, and paved road dust represents the largest source of particle-associated metal emissions to the atmosphere. In order to better understand the role of roadways as a source of localized metal deposition, we characterized the horizontal dry deposition patterns of chromium, copper, lead, nickel and zinc upwind and at increasing distances downwind of the I-405 Freeway in coastal Los Angeles. Dry deposition fluxes and atmospheric concentrations of these metals were highest at the site closest to the freeway, and reduced to approximately urban background concentrations between 10 and 150 m downwind of the freeway. Compared with urban background, atmospheric particle size distributions indicated the freeway was a significant source of these metals on large particles > 6 µm in diameter, which deposit close to their source and account for the increased dry deposition flux rates observed near the freeway. The spatial pattern of measured deposition flux was well predicted by a relatively simple line-source Gaussian plume model modified to include particle deposition and resuspension. The model results indicated dilution by vertical dispersion of the plume was the most important mechanism regulating downwind concentrations and deposition. © 2006 Elsevier Ltd. All rights reserved.

**Keywords:** Atmospheric deposition; Large particles; Urban runoff; Paved road dust

### 1. Introduction

Dry atmospheric deposition near urban centers, especially in semi-arid regions such as Southern California, represents a potentially important non-point source of particle-associated metals to waterbodies (Baker et al., 1997; Lu et al., 2003). Atmospheric particulate matter may be directly

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deposited onto the surface of a waterbody or may reach the waterbody indirectly through deposition onto the land surface during dry periods, followed by subsequent wash-off during storm events. Atmospheric deposition may be particularly important in the Los Angeles Air Basin, since air quality in this region, with a population greater than 17 million, ranks among the worst in the United States (SCAQMD, 2000). Emission inventories of the basin indicate significant quantities of toxic materials are regularly released into the atmosphere (SCAQMD, 2003), and the ultimate fate of the heavy metals in particular is unknown.

In urban areas, emissions from paved roadways are a major source of atmospheric particulate matter (Dunbar, 1976; Cowherd et al., 1979; Reider, 1983; Cowherd and Englehart, 1984, 1985). Investigation of the most recent emission inventories for the Los Angeles region by Stolzenbach et al. (2003) found resuspended dust represents the largest source of particle-bound pollutant metals in the Los Angeles region, with paved road dust representing the most significant fraction. Paved road dust originates from pavement wear and decomposition, dustfall, litter, mud and dirt carryout, spills, biological debris, and erosion from adjacent areas (Cowherd and Englehart, 1984; Chow et al., 1990; Chow and Watson, 1992). In an urban setting, a source of zinc and copper in paved road dust is from vehicle exhaust and vehicle brake and tire wear (Watson et al., 2000; Councell et al., 2004).

Because air quality standards have been set for particles less than  $10\mu\text{m}$  in diameter due to human health concerns, most research on road dust emissions has focused on particles in this size fraction (Cowherd and Englehart, 1984; Kantamneni et al., 1996; Venkatram and Fitz, 1998; Fitz, 2001). However, the results from a number of studies indicate nearly 50% of road dust total suspended particulate matter (TSP) emissions are due to particles larger than  $10\mu\text{m}$  (Ahuja et al., 1989; Houck et al., 1989, 1990). Moreover, particles greater than  $10\mu\text{m}$  in diameter are largely responsible for metal deposition (Lin et al., 1993, 1994; Paode et al., 1998; Zufall et al., 1998). Because coarse particles settle faster due to their greater inertia and gravitational settling, deposition of these particles is likely to occur relatively close to their source (Schmel, 1973). Previous studies have documented a pattern of locally high atmospheric concentrations of particulate matter near roadways using tracers and downwind direct measurements of

air and ground surface concentrations (Claiborn et al., 1995; Hitchins et al., 2000; Zhu et al., 2002a,b); however, few studies have focused on particle deposition gradients near roadways, especially for particles larger than  $10\mu\text{m}$ .

In addition to deposition as a source of metal loading into water bodies, re-entrainment of suspended atmospheric particles contributes to the dispersion of pollutants and impacts the subsequent mass loading into water bodies. The size of particles that can be easily resuspended ranges from  $1\mu\text{m}$  to  $50\mu\text{m}$  in diameter. Resuspended particles are estimated to travel globally; for example, Asian dust has been identified in Hawaii (Parrington et al., 1983) and Sahara dust in the central US (Perry et al., 1997). According to Sternbeck et al. (2002), measured metal concentrations in air are generally similar to the chemical profiles of crustal elements. This result indicates resuspension may control particle abundance and chemical composition.

The re-entrainment and suspension of particles in the atmosphere may occur through several natural and anthropogenic processes. Meteorological conditions during and after deposition (e.g. wind speed and intensity of rain) and surface characteristics, such as surface roughness and surface moisture, are important influences on natural resuspension (Nicholson, 1988). Anthropogenic activities such as vehicular activities, agricultural activities and various cleaning operations induce resuspension (Kashparov et al., 1994; Garger et al., 1998).

Because of the difficulty of measuring concentrations under different atmospheric stability and roadway configurations, predictions of concentration and dispersion of particulate matter near roadways have been made using line-source Gaussian plume models (Chock, 1978; Horst, 1978; Sisla et al., 1979). The most widely used versions of these models are the modified HIWAY (Zimmerman and Thompson, 1975), HIWAY-2 (Peterson, 1980), GM (Chock, 1978), and CALINE-3 (Benson, 1979). However, these models have primarily been used to estimate vapor phase concentrations of constituents such as carbon monoxide, and do not include deposition or resuspension.

This study was designed to gain a better understanding of the role of major roadways, such as a freeway, as a significant source of localized metal deposition to urban surfaces, and to understand the role of resuspension in the net deposition and dispersion of particulate matter near roadways. To accomplish this goal, the following objectives were

defined: (1) to characterize the horizontal dry deposition gradient and atmospheric concentrations of five pollutant metals (chromium, copper, lead, nickel and zinc) near a major freeway in Los Angeles; (2) to compare atmospheric particle size distributions of these metals near a freeway with urban background values; and (3) to compare measured horizontal deposition fluxes with the predictions of a Gaussian line-source dispersion model modified to include deposition and resuspension.

## 2. Methods

### 2.1. Field sampling

#### 2.1.1. Site description

The freeway site selected for this study was the I-405 freeway between Wilshire and Sunset Boulevards in West Los Angeles (Fig. 1). This site provided appropriate conditions to study the impact of the freeway on localized deposition for two reasons. First, this stretch of freeway runs in a north-west to south-east direction, and is approximately perpendicular to the on-shore, southwest winds which are typically dominant in coastal Los Angeles during daytime in the spring. Second, the I-405 freeway has a heavy traffic volume, with an annual average daily traffic count of approximately 300,000 (California Department of Transportation and (CA DOT), 2004). All sampling equipment was located along Constitution Blvd, which runs perpendicular to the freeway. This site has been used successfully by previous researchers documenting

air pollutant concentration gradients with distance from the freeway (Zhu et al., 2002b).

Dry deposition flux of metals was measured simultaneously at three sites located at 10 (DW1), 150 (DW2) and 450 m (DW3) downwind, and one site approximately 150 m upwind (UP) of the I-405 Freeway over a three-week period in April and May, 2003. All deposition measurements were collected during daytime, high traffic hours (e.g. from approximately 8 AM to 5 PM) over a period of three to four days for a single sample, in order to obtain sufficient mass. Simultaneously with the dry deposition flux measurements, atmospheric concentrations of TSP were collected at two downwind sites (DW1 and DW3) and analyzed for both particle mass and metal concentrations. Atmospheric concentrations of metals on four coarse particle size fractions were also measured at the downwind site closest to the freeway (DW1) one day each week. All sampling took place during typical meteorological conditions for spring in Los Angeles, during periods with no precipitation. Local wind data were utilized during sampling to confirm the upwind and downwind status of the sampling locations.

#### 2.1.2. Instrumentation

Dry deposition flux measurements were made using a 33 cm diameter circular polyvinyl chloride (PCV) plate with a sharp edge ( $<10^\circ$  angle), covered with a Mylar® sheet coated with Apiezon L grease. The grease was liquefied by heating and then painted onto the Mylar film to obtain a thin, uniform 10  $\mu\text{m}$  layer. This surrogate surface has been used previously and is described in more detail elsewhere (Lim et al., 2006). During sampling, the plate was mounted onto a tripod at a height of 2 m. A filter-based sampling system attached to a vacuum pump was used to collect TSP for measurement of atmospheric concentrations of metals. The open-faced inlet was loaded with a 37 mm, 2.0  $\mu\text{m}$  pore Teflon® filter (Pall Life Science), and sampling was done at a flow rate of 10 L min<sup>-1</sup>. Meteorological data, including wind speed and direction, temperature, relative humidity and barometric pressure, were also measured during each sampling period using a portable meteorological station (PortLog, Rain Wise, Inc.) located at the DW2 site.

A Noll Rotary Impactor (NRI) was used to collect size distributions of particulate matter on different coarse particle fractions for measurement

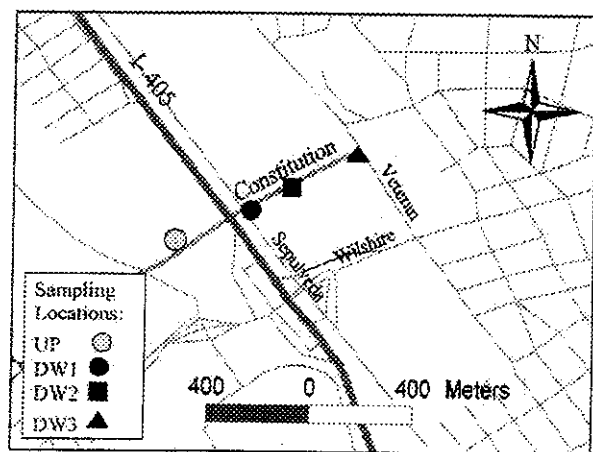


Fig. 1. Map of the freeway site and sampling locations.

of atmospheric concentrations of metals. This instrument has been used successfully to measure air concentrations on coarse particle size fractions in other studies (Mamane and Noll, 1985; Noll et al., 1985; Lin et al., 1994). The NRI operates by simultaneously rotating four rectangular collector stages through the air at high velocity to collect particles directly from ambient air by impaction. Each collector stage is a different width to collect a different particle size fraction. The collector stages were mounted with Mylar strips, sized according to the size of the collector stage, which were coated with Apiezon L grease in the same manner as the Mylar for the deposition plates. The instrument was operated at 320 rpm, producing cut diameters of 6, 11, 20 and 29  $\mu\text{m}$  for the four collector stages. To prevent overloading, the smallest collector stage (stage A) was changed every 2 h, while the next largest collector stage (stage B) was changed every 4 h. The two largest collector stages (stages C and D) were not changed during the ~8-h collection period. The metal concentration on the particle size fraction smaller than 6  $\mu\text{m}$  was obtained by subtracting the metal mass concentration collected on the NRI Stage with the cut diameter of 6  $\mu\text{m}$  from the TSP metal mass concentration.

#### 2.1.3. Sample handling and chemical analysis

Prior to sampling, Mylar was cut to the desired size (33 cm diameter circles for the deposition plates and rectangular strips sized for the desired cut points of the NRI) and cleaned by wiping with methanol-soaked wipes, then immersed in 10% nitric acid followed by methanol for five minutes each. After cleaning, the Mylar pieces were rinsed with distilled water and allowed to air dry. Dry Mylar strips and circles were coated with a thin layer of Apiezon L grease and mounted onto the collector stages of the NRI and the deposition plates, respectively, one day prior to sampling. Both the NRI stages and the deposition plates were stored in airtight containers for transport to the field.

After sampling, the Mylar strips were removed from the NRI stages and each strip stored in a clean Petri dish. The Mylar circles were removed from the deposition plates, folded (greased side inward), and placed inside a clean glass jar. In the lab, Mylar circles were divided into 10 smaller pieces, and the 10 pieces returned to the original sample jars. The Mylar strips were placed into clean 15 mL plastic centrifuge tubes. The Mylar circles and strips were

rinsed three successive times with 15 mL of *n*-hexane. The rinses were combined into a 50 mL centrifuge tube. The Mylar pieces were then rinsed with 5% Optima Grade nitric acid and the acid rinses were added to the combined hexane rinses. The hexane was evaporated in a 50 °C water bath and the remaining acidified sample was then heated to 65 °C under sonication for a minimum of 24 h.

For the TSP instrument, a clean Teflon filter was weighed after equilibrating for 24 h in a room with approximately constant temperature and humidity conditions. After weighing, the filter was loaded into the sample holder, and the sample holder stored in a clean, airtight plastic bag for transport to the field. After sampling, the filter was removed from the holder and placed in a clean petri dish and transported to the lab. Prior to analysis, the filters were again weighed. The filters were transferred into clean 15 mL plastic centrifuge tubes, and 10 mL of 5% Optima Grade nitric acid were added and the tubes capped tightly. The acidified samples were then heated to 65 °C under sonication for a minimum of 24 h.

All acid-digested samples were transferred to a centrifuge tube and analyzed for 26 metals per EPA Method 200.8 using inductively coupled plasma-mass spectroscopy (ICP-MS). Results reported here are for chromium, copper, lead, nickel and zinc, because these are the primary metals associated with water quality issues in Southern California. Method detection limits were 0.5 ng for lead and 1 ng for all other metals, corresponding to a minimum detectable air concentration of 0.02  $\text{ng m}^{-3}$  and a minimum detectable deposition flux of 0.004  $\mu\text{g m}^{-2} \text{day}^{-1}$  for lead and 0.01  $\mu\text{g m}^{-2} \text{day}^{-1}$  for all other metals. Laboratory blanks were analyzed with each batch of 15 samples. In addition to samples, field blanks (greased Mylar sheets mounted onto a duplicate deposition plate, Teflon filters loaded into a TSP sampling cartridge) were prepared, taken to the field during a sampling event, and analyzed along with the samples collected at the site. All laboratory blanks were non-detects; however all field blanks contained detectable levels of metals, and all samples were corrected for their respective field blank.

Field duplicates indicated the precision of the deposition plates. For each metal, the relative percent difference (RPD) between duplicate deposition plates, on average, was 31% (chromium), 25% (copper), 87% (nickel), 24% (lead), and 47% (zinc). This was an acceptable level of precision for field duplicates because differences of less than a factor of two

between fluxes measured during different sampling events were typically not considered significant.

#### 2.1.4. Image analysis of NRI Mylar strips

An image processing program was used to count particles and to obtain the particle size and mass distributions of the particles deposited on the NRI stage A ( $d_p > 6 \mu\text{m}$ ) from photographs of the Mylar strips taken with an optical microscope (LW Scientific) set at a magnification of  $100\times$ . For each image the distribution of the aerodynamic particle diameter  $d_p$  was determined using

$$d_p = \frac{1}{S_v} \left( \frac{\rho_p}{\rho_0 S_D} \right)^{1/2} d_{PA}, \quad (1)$$

where  $d_{PA}$  is the equivalent projected area diameter measured by the image analysis,  $S_D$  is a dynamic shape factor set equal to 1.41 (Davies, 1979),  $\rho_p$  is the particle density assumed to be  $1800 \text{ kg m}^{-3}$ ,  $\rho_0$  is a unit particle density of  $1000 \text{ kg m}^{-3}$ , and  $S_v$  is the volume averaged shape factor set equal to 1.61 for urban sites (Lin et al., 1994; Tai et al., 1999). The aerodynamic diameter ( $d_p$ ) and assumed particle density ( $\rho_p$ ) are then used to calculate total particle volume and mass. The atmospheric concentration is obtained using the known NRI rotation speed and empirically determined collection efficiencies (Noll et al., 1985).

#### 2.2. Modeling approach

The model used in this study was based on the ground-level line-source Gaussian plume model, formulated to consider metal deposition and resuspension (Horst, 1978). Assumptions for the model were constant emission rate, constant wind speed both in time and space, neutral stability (stability D), and a flat and unobstructed ground surface (Masters, 1998). The emissions from the freeway were assumed to form a single, continuously emitting, infinite line source with metal mass flow per unit length  $q_0$  ( $\text{mg m}^{-1} \text{ s}$ ).

With these assumptions, the metal concentration in the air down wind at distance  $x$  and elevation  $z$  from the line source can be described by the following:

$$C(x, z) = \frac{1}{u\sqrt{\pi/2}} \left\{ \frac{q_0}{\sigma_z(x)} \exp\left(\frac{-z^2}{2\sigma_z^2(x)}\right) + \int_0^x \frac{m(\xi)}{\sigma_z(x-\xi)} \exp\left(\frac{-z^2}{2\sigma_z^2(x-\xi)}\right) d\xi \right\}, \quad (2)$$

where  $u$  is the wind speed;  $m(x)$  = net metal mass flow per unit ground surface area ( $\text{mg m}^{-2} \text{ s}^{-1}$ ) at a distance  $x$  resulting from deposition and resuspension;  $\sigma_z(x) = c(x)^d + f$  = vertical standard deviation of the plume at a distance  $x$  from the plume source, where  $c$ ,  $d$ , and  $f$  are constants that are a function of the stability classification (Masters, 1998).

The net metal mass flow to the atmosphere per unit ground surface area resulting from deposition and resuspension is computed by

$$m(x) = \Lambda G(x) - V_d C(x, 0), \quad (3)$$

where  $G(x)$  ( $\text{mg m}^{-2}$ ) is the surface metal mass per unit area,  $\Lambda$  ( $\text{s}^{-1}$ ) is a specified resuspension rate,  $C(x, 0)$  is the ground level metal concentration in the air, and  $V_d$  is a specified deposition velocity. The change in surface contamination with time is then given by

$$\frac{dG(x)}{dt} = -m(x) = V_d C(x, 0) - \Lambda G(x), \quad G(x) = 0 \quad \text{at } t = 0. \quad (4)$$

The build-up of  $G(x)$  is the only time dependent process in the model, although changes in  $G(x)$  drive changes in all other variables. A steady-state condition where  $m(x) = 0$  and  $\Lambda G(x) \approx V_d C(x, 0)$  is reached in a time of about  $1/\Lambda$ . The steady-state atmospheric metal concentration is given by the first term in Eq. (2) and is independent of the deposition velocity  $V_d$  and the resuspension rate  $\Lambda$ , but the deposition flux is  $V_d C(x, 0)$ .

### 3. Results and discussion

#### 3.1. Meteorological data

There was little day-to-day variability in the meteorological data measured during the sampling at the freeway site (Table 1), and even hour-to-hour variability during the 8 AM–5 PM sampling period within the same day was low. Because meteorological conditions were stable throughout the study period, we did not attempt to correlate these data with our weekly concentration or deposition flux measurements. Wind direction remained predominately from the southwest on all sampling days, as expected for springtime in Los Angeles, maintaining the desired upwind and downwind locations of our sampling sites.

Table 1  
Summary of mean meteorological data measured during sampling at the freeway site

Week	Date	Temperature (°C)	Relative humidity (%)	Wind speed (m s <sup>-1</sup> )		Wind direction from:
				Mean 8-h	Range of 10-min Max	
1	13-April-04	19±1	63±4	2.1±0.6	0.4–7.2	Southwest
	14-April-04	19±1	62±4	2.1±0.7	2.2–7.2	Southwest
	15-April-04	20±1	62±2	2.1±0.7	2.7–6.7	Southwest
	16-April-04	18±1	61±5	2.1±0.6	3.1–6.7	Southwest
2	19-April-04	18±1	54±4	2.4±0.8	2.7–8.0	Southwest
	20-April-04	19±1	60±5	2.4±0.7	2.7–7.2	Southwest
	21-April-04	19±1	64±3	2.4±0.6	3.1–7.2	Southwest
3	28-April-04	21±2	67±7	2.2±0.5	2.7–7.2	Southwest
	29-April-04	20±1	55±4	2.4±0.5	3.1–7.2	Southwest
	30-April-04	19±1	67±5	2.7±0.6	3.1–7.2	Southwest
	1-May-04	26±1	38±5	2.5±0.4	4.0–6.3	Southwest

Ten-minute data were recorded. The 8-h means are presented here, except as noted.

Table 2  
Mean dry deposition flux±standard deviation (μg m<sup>-2</sup> day<sup>-1</sup>) of metals measured at varying distances from the I-405 Freeway

Location	Chromium	Copper	Nickel	Lead	Zinc
10 m Downwind (DW1)	4.3±0.3	48±8	3.1±0.6	24±3	144±33
150 m Downwind (DW2)	2.4±1.9	18±7	1.0±1.3	11±4	45±23
450 m Downwind (DW3)	2.5±1.1	14±3	1.2±1.1	7.3±1.8	38±1
150 m Upwind (UP)	2.2±0.5	11±5	1.5±0.9	7.9±1.3	37±10

N = 3 for all metals.

### 3.2. Dry deposition flux gradient near freeway

The highest measured deposition fluxes of metals were observed at the downwind site closest to the freeway (DW1) for all five metals, although the largest differences were observed for copper, lead and zinc (Table 2). For copper, lead and zinc, ANOVA indicated the difference between sites was significant ( $p < 0.002$ ). This difference was due entirely to the higher fluxes observed at the DW1 site, based on the Tukey test for pairwise multiple comparisons ( $p < 0.003$ ). Mean fluxes at DW1 were higher than UP and DW3 by factors of two to five, depending on the metal. Mean fluxes at DW1 were higher than DW2 by factors of two to three for all five metals. In contrast, for all five metals, differences in fluxes measured at the UP site (considered to represent "urban background"), DW2, and DW3 sites were not significant (ANOVA,  $p > 0.05$ ).

These results indicated dry deposition fluxes of metals were higher at short distances from the freeway, and quickly reduced to urban background

fluxes within 150 m, especially for copper, lead and zinc. These results were similar to the observations of Zhu et al. (2002b) for ultra fine particle concentrations ( $d_p < 0.1 \mu\text{m}$ ) measured downwind of the same freeway in which high concentrations near the freeway reduced to urban background within 300 m. The small sample size in this study was an important limitation of these data. However, the general trend of decreasing deposition flux with distance from the freeway was consistently observed during the study period. These results also suggested sources of deposited copper, lead and zinc may be different from chromium and nickel. The fluxes of copper, lead and zinc observed to be significantly higher close to the freeway suggest (1) the freeway acts as a significant source of these metals and (2) these metals had substantial concentrations on larger particles, which are expected to deposit close to their source. The freeway likely represents a source of large particles containing copper, lead and zinc because of resuspension of road dust as vehicles travel on the freeway at high velocities (Sehmel, 1973; Nicholson et al., 1989), and from tire and



brake wear from vehicles (Chow et al., 1990; Chow and Watson, 1992; Councell et al., 2004).

### 3.3. Atmospheric concentrations near the freeway

TSP metal concentrations were higher at the downwind site closest to the freeway (Fig. 2a), although the differences between DW1 and DW3 were only significant for lead and zinc using the paired samples *t*-test ( $p < 0.02$ ). TSP particle mass concentrations were also consistently higher at the downwind site closest to the freeway (DW1) compared with the furthest site (DW3) (Fig. 2b), however differences between the sites were not statistically significant. Again, the sample size ( $n = 3$ ) may have limited our ability to detect small differences between locations.

### 3.4. Relative distribution of metals on different particle size fractions

The relative distribution of metal mass among particle size fractions, as determined from the NRI and TSP metal concentration data, was different near the freeway from distributions observed at

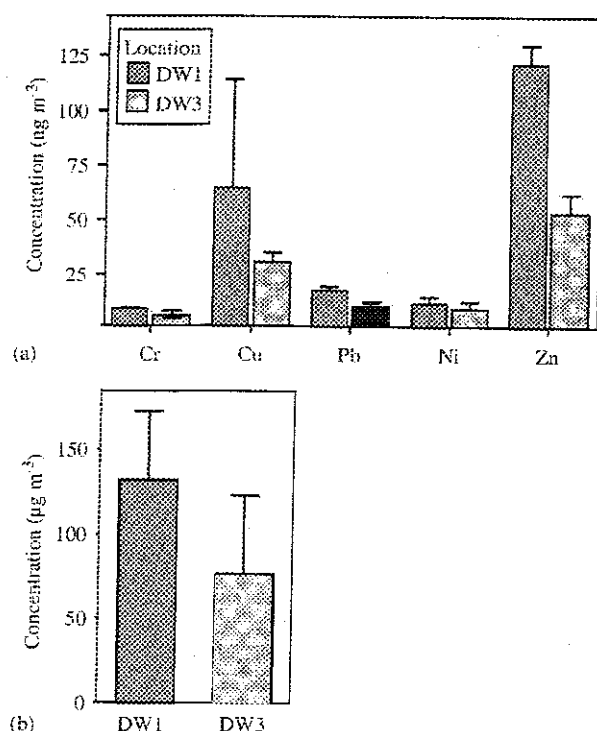


Fig. 2. Mean atmospheric TSP concentrations downwind from the I-405 Freeway of (a) metals and (b) particulate matter mass. DW1 = 10m downwind; DW3 = 450m downwind. Error bars represent the standard deviation for the mean.

urban background locations. Fig. 3 shows the mean percent contribution to five particle size fractions ( $< 6$ , 6–11, 11–20, 20–29, and  $> 29 \mu\text{m}$ ) for each metal measured near the freeway, and compared with the mean distribution measured at urban background sites in Los Angeles located away from major freeways using the same instrumentation (Lim et al., 2006). The primary difference between the freeway site and the urban background sites was that more metal mass fraction was observed in the particle fraction  $> 6 \mu\text{m}$  at the freeway site, especially for copper, lead, and zinc, while at the urban background sites, the majority of the mass fraction (approximately 75%) for all metals was observed in the particle size fraction  $< 6 \mu\text{m}$ .

One limitation of this study is that we did not have simultaneous measurements of the particle size distribution near the freeway and at urban background. However, as we observed both lower TSP metal concentrations at the DW3 site, and reduced metal mass fraction due to the largest particles at urban background locations away from the freeway, we hypothesize the lower metal concentrations were due to removal of the largest particles by deposition near their source. Thus, we conclude differences in the particle size distributions of metal mass at the freeway site compared with urban background sites likely

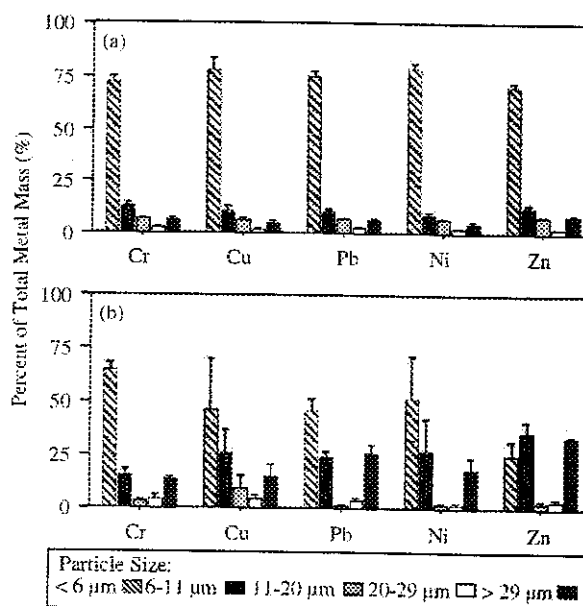


Fig. 3. Mean distributions of metal concentrations on five particle size fractions as a percent of the total metal mass measured (a) at urban background sites as measured by Lim et al. (2006) and (b) 10m downwind of the I-405 Freeway. Error bars are the standard error of the mean.

resulted from the freeway acting as an emission source for particles  $>6\mu\text{m}$ . These coarse particles are more readily removed from the air by deposition close to the source, and thus we expect to find them at lower concentrations, and as a lower percentage of the total mass, as distance from the emission source increases. Smaller particles ( $<6\mu\text{m}$ ), which are slow to deposit, remain suspended and contributed a higher proportion of the total mass away from local sources. This explains why both atmospheric concentrations of metals on the largest particles and metal deposition fluxes were reduced at sites further removed from a freeway. Corroborating results from our deposition plate data indicated the majority of this removal occurred very near to the source (e.g., between 10 and 150 m), and resulted in deposition fluxes and concentrations downwind of the freeway comparable to urban background at some distance greater than 10 m, and within 150 m.

### 3.5. Image analysis results

Size-segregated hourly variation of particulate mass concentration for particles greater than  $6\mu\text{m}$  in diameter was estimated using image analysis of the stage A NRI strips. A comparison of the particulate mass concentration at an urban background site (as measured by Lim et al., 2006) and the freeway transect is shown in Fig. 4. The particle mass observed in the fraction  $>29\mu\text{m}$  at the freeway site was relatively constant over time, while this size fraction was absent during the morning period, 7:00–11:00 AM, at the urban background site, which had higher particulate matter concentrations only later in the day, probably as a cumulative result of resuspension from traffic. The total particle mass concentration associated with particles  $>6\mu\text{m}$  at the freeway site ranged from  $23\text{--}31\mu\text{g m}^{-3}$ , compared with  $10\text{--}16\mu\text{g m}^{-3}$  at the urban background site. The higher particle mass concentration at the freeway site most likely reflected the close proximity to the freeway. Using ANOVA ( $p>0.5$ ), the time variation of the total particulate mass concentration at the freeway site for particles  $>6\mu\text{m}$  did not show any significant difference between times, likely because of the relatively constant traffic flow on this particular freeway.

### 3.6. Modeling results

Model calculations used a wind speed  $u = 2\text{ m s}^{-1}$ , which was the average value during the

observation period, and a deposition velocity  $V_d = 0.01\text{ m s}^{-1}$ , which was the mean of the flux-averaged deposition velocities calculated for each metal by dividing the measured deposition flux by the air concentration measured by the TSP sampler at both the DW1 and DW3 locations (Table 3). Resuspension rates  $\lambda$  reported in the literature for surfaces of all types vary from  $10^{-13}$  to  $10^{-6}\text{ s}^{-1}$  (Nicholson, 1988), and for asphalt surfaces from  $5 \times 10^{-9}$  to  $6 \times 10^{-8}\text{ s}^{-1}$  (Sehmel, 1980). For the present study, values of  $\lambda$  between  $10^{-9}$  and  $10^{-6}\text{ s}^{-1}$  reflect the possible range of surface conditions, corresponding to a time to steady-state ranging from 10 to 10,000 days and indicating the surface metal concentration reflects an accumulated average of time-varying conditions. For this reason, the deposition flux measurements were compared with the deposition flux predicted by the steady-state model solution. Because the steady-state atmospheric concentration distribution reflects a balance between deposition and resuspension, the calculated atmospheric concentration and deposition flux are independent of the assumed resuspension rate. Calculations with  $\lambda = 0$ , i.e. no resuspension but net loss by deposition, provided an upper bound on the effect of losses by deposition on the atmospheric concentration and associated deposition flux.

Steady-state model predictions of downwind deposition patterns were compared with observed values of normalized “excess” deposition, calculated by subtracting the calculated and measured deposition rates at the upwind site, which are zero in the model, from the calculated and measured values at the downwind sites and dividing the difference by the excess value calculated and observed at the closest downwind site. This normalization, which was necessary because the line source emission rate  $q_0$  was unknown, removes any effect of the absolute magnitude of deposition and focuses on the downwind spatial variation of deposition.

Good agreement between the steady-state model calculation and measured deposition fluxes was obtained for chromium, copper, lead, and zinc, particularly within 200 m from the source (Fig. 5). The data for nickel diverged from the model results, which was not surprising because of the expected absence of a nickel source at the freeway. Both the model and the measurement data indicated deposition and air concentrations return to background between 100 and 150 m from the source; a result consistent with the findings of Zhu et al. (2002b) for

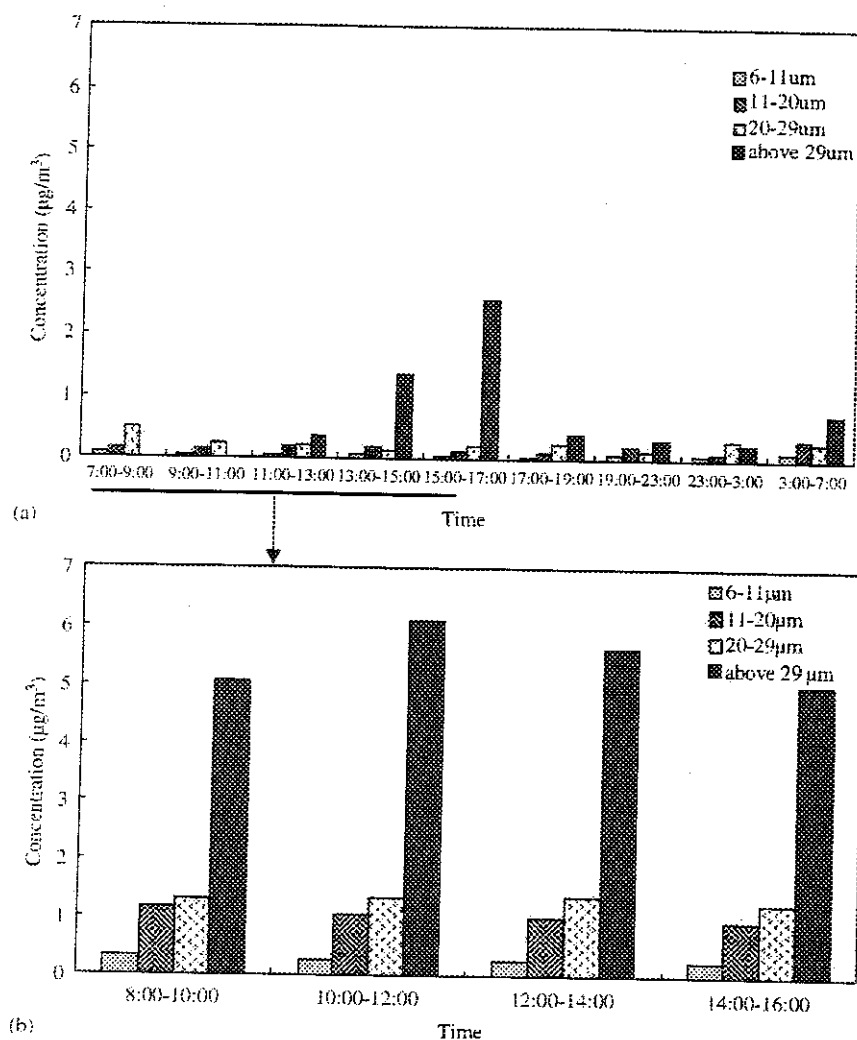


Fig. 4. Diurnal pattern of particulate mass concentration measured using image analysis: (a) urban background site as measured by Lim et al. (2006); (b) freeway site (this study).

Table 3  
Measured flux-averaged deposition velocities for different metals

Metal	N	Deposition velocity ( $\text{cm s}^{-1}$ )
Chromium	6	$0.56 \pm 0.1$
Copper	6	$1.4 \pm 2$
Lead	6	$1.3 \pm 0.5$
Nickel	6	$0.26 \pm 0.2$
Zinc	6	$1.1 \pm 0.4$

ultra fine particles. Model calculations with  $\Lambda = 0$  were nearly identical to the steady-state solution, indicating that on the scale of this experiment

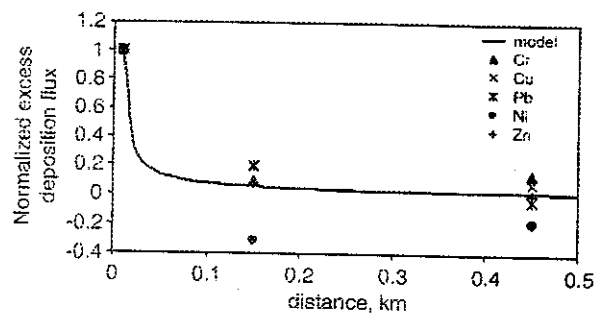


Fig. 5. Comparison of computed and measured normalized excess deposition flux (see text for definition). The solid line is the steady-state model prediction using a wind speed  $u = 2 \text{ m s}^{-1}$  and a deposition velocity  $V_d = 0.01 \text{ m s}^{-1}$ .

atmospheric dispersion is the major mechanism determining the spatial distribution of atmospheric metal concentration and metal deposition flux.

#### 4. Conclusions

Our findings indicated metal deposition was increased in the immediate vicinity of a large freeway, and quickly reduced to urban background deposition rates between 10 and 150 m downwind of the freeway, especially for copper, lead and zinc. Furthermore, higher concentrations at the downwind site closest to the freeway were primarily due to increased mass in the particle size fraction  $>6\mu\text{m}$  because the freeway acts as an emission source for particles  $>6\mu\text{m}$ , which are removed from the air by deposition close to the source. Because of resuspension, particles  $>6\mu\text{m}$  are consistently observed at urban background locations, but as a smaller percentage of the total mass as distance from the emission source increases. These results suggest (1) the freeway acts as a significant source of copper, lead and zinc and (2) these metals have substantial concentrations on larger particles emitted from the freeway due to the dispersion of road dust by vehicles traveling at high speeds. Furthermore, the modified Gaussian plume model showed relatively good agreement with deposition measurements, dispersion being the most significant process controlling the spatial variation of concentration and deposition.

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## **Mercury concentrations in coastal California precipitation: Evidence of local and trans-Pacific fluxes of mercury to North America**

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### **Abstract**

Because of mercury's (Hg) relatively high vapor pressure and long (0.5–2 years) atmospheric residence, there is the potential for long-range transport of contaminant Hg. Many studies have focused on that transport and deposition in central and eastern North America, Europe, and the Arctic, but there has been little research on the cycling of Hg in the western coast of North America. That deficiency is addressed in this preliminary study, which indicates there is long-range transport of Hg across the North Pacific. This transport is evidenced by the elevated (relative to equatorial and theoretical baseline) Hg concentrations in rainwater collected on the coast of California, as well as by the positive correlation between North Pacific storm tracks and Hg concentrations, with maximum concentrations associated with storms from 20°–40° latitude. Those tracks trace air masses containing industrial emissions with peak O<sub>3</sub> concentrations moving eastward off the Asian continent. The Asian fluxes appear to enhance Hg concentrations both directly, through the emission of particle-bound Hg and reactive Hg<sup>2+</sup>, and indirectly, by increasing the rate of oxidation of Hg<sup>0</sup> in the atmosphere. Superimposed on the trans-Pacific background of industrial Hg is a local signal, with elevated concentrations at the urban site relative to the more pristine coastal site in California. This secondary enrichment is tentatively attributed to elevated local emissions of redox species, including O<sub>3</sub> and its precursors, which increase oxidation rates of Hg<sup>0</sup> in the atmosphere and Hg concentrations in precipitation.

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**Index Terms:** 0365 Atmospheric Composition and Structure: Troposphere—composition and chemistry; 0368 Atmospheric Composition and Structure: Troposphere—constituent transport and chemistry; 0345 Atmospheric Composition and Structure: Pollution—urban and regional (0305).

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